

Effect of the Ba/Sr ratio on the optical properties of phosphate laser glass

*A.S.Popov*¹, *A.V.Uklein*¹, *A.N.Zaderko*²,
*V.A.Kozhanov*², *V.V.Lisnyak*², *V.Ya.Gayvoronsky*¹

¹Institute of Physics, National Academy of Sciences of Ukraine,
46 Nauki Ave., 03680 Kyiv, Ukraine

²Chemical Faculty, T.Shevchenko National University of Kyiv,
62a Volodymyrska Str., 01601 Kyiv, Ukraine

Received March 16, 2015

Nonlinear optical (NLO) approach was utilized for the laser phosphate glasses characterization, being prepared by the molar substitution of Ba²⁺ for Sr²⁺ at a fixed ratio of other components in a mixture of Ba_{1-x}O/Sr_xO–P₂O₅–Al₂O₃–B₂O₃–RE₂O₃ (RE : Nd, La). The optical characterization of the glasses was provided by optical and FTIR spectroscopy and elastic scattering indicatrices analyses. It was shown that an efficiency of the photoinduced absorption/bleaching depends on the substitution Ba²⁺–Sr²⁺ level x due to the self-action of pulsed and CW laser radiation at wavelength 1064 nm. The proposed technique can be applied for the oxygen stoichiometry diagnostics for laser phosphate glasses. For the $x = 0.25$ composition it was obtained UV-vis range absorption and elastic scattering reduction within the lowest efficiencies of the NLO absorptive response under pulsed and CW laser excitation at wavelength 1064 nm. Such optical properties improvement was achieved by both the substitution level choice modification and the purification of the initial reagents.

Keywords: Laser phosphate glasses, Ba/Sr ratio, photoinduced absorption, cubic nonlinear optical susceptibility, UV-visible and FTIR spectra, elastic scattering indicatrix.

Проведена характеристика нелинейно-оптического (НЛО) отклика фосфатного лазерного стекла, изготовленного путем замещения ионов Ba²⁺ на Sr²⁺ при фиксированном составе других компонент в растворе замещения Ba_{1-x}O/Sr_xO–P₂O₅–Al₂O₃–B₂O₃–RE₂O₃ (RE : Nd, La), совместно со спектральным анализом в видимом и ИК диапазонах и исследованием индикатрис упругого рассеяния света. Показано, что эффективность процессов фотоиндуцированного поглощения/просветления зависит от степени замещения Ba²⁺–Sr²⁺– x при самовоздействии непрерывного и импульсного лазерного излучения с длиной волны 1064 нм. Предложенная методика может быть применена для диагностики стехиометрии по кислороду в фосфатных лазерных стеклах. Для состава с $x = 0.25$ наблюдалось уменьшение спектрального поглощения и упругого рассеяния света при минимальной эффективности абсорбционного НЛО отклика в условиях импульсного и непрерывного лазерного возбуждения на длине волны 1064 нм. В результате получено улучшение оптических свойств фосфатного стекла путем подбора степени замещения и очистки первичных реагентов.

Вплив вмісту Ba/Sr на оптичні властивості фосфатного лазерного скла. *О.С.Попов, А.В.Уклеїн, А.М.Задерко, В.О.Кожанов, В.В.Лісняк, В.Я.Гайворонський.*

Проведено характеристику нелінійно-оптичного (НЛО) відгуку фосфатного лазерного скла, що було виготовлено шляхом заміщення іонів Ba²⁺ на Sr²⁺ при фіксованому вмісті інших компонент у розчині заміщення Ba_{1-x}O/Sr_xO–P₂O₅–Al₂O₃–B₂O₃–RE₂O₃ (RE : Nd, La), разом із спектральним аналізом у видимому та ІЧ діапазонах та

дослідженням індикатрис пружного розсіювання світла. Показано, що ефективність процесів фотоіндукованого поглинання/просвітлення залежить від ступеня заміщення $Ba^{2+}-Sr^{2+} - x$ при самовпливі неперервного та імпульсного лазерного випромінювання з довжиною хвилі 1064 нм. Запропонована методика може бути застосована для діагностики стехіометрії за киснем у фосфатному лазерному склі. Для складу $x = 0.25$ спостерігалось зменшення спектрального поглинання та пружного розсіювання світла разом з мінімальною ефективністю абсорбційного НЛО відгуку при імпульсному та неперервному лазерному збудженні на довжині хвилі 1064 нм. В результаті отримано покращення оптичних властивостей фосфатного скла шляхом підбору ступеня заміщення та очищення первинних реагентів.

1. Introduction

Glassy materials with a high quantum yield of the luminescence and improved thermo-optical characteristics are used for the production of large-sized workpieces for the laser application. These optical workpieces show a low crystallization propensity and have reduced efficiency of the refractive nonlinear response. A number of phosphate-based glasses of such a type are known for nowadays [1–7]. Among them are the optical phosphate glasses that composed of (in mass%) 55–65 P_2O_5 , 8–12 BaO, 10–15 K_2O , 1–4 SiO_2 , 5–10 Al_2O_3 , 2–6 B_2O_3 , 0.1–4.5 Nd_2O_3 . These optical glasses have high values of stimulated transitions cross sections, of about $3.5 \cdot 10^{-20} \text{ cm}^2$, and high luminescence decay time. However, they demonstrate a low absorption at the wavelength of 1053 nm, which magnitude is less than 0.001 cm^{-1} [3]. The improvement of the glasses characteristics can be reached with: (i) the reduction of concentration quenching of luminescence, which depends on the distances between Nd atoms in a glass matrix; (ii) reduction of the photoinduced absorption efficiency [7] at high radiation densities; (iii) decreasing the propensity to crystallization and (iv) lowering the content of colouring impurities. These changes could be done, at the stage of the glass preparation, through convenient chemical methods. These methods direct on the changing the composition of these glasses and the purity of raw materials that are used at the glass production stage.

There two types of substitution can be performed in such glasses: (1) the substitution of Ba by Sr ions that controls oxygen stoichiometry and (2) Nd^{3+} by La^{3+} ions that reduces Nd_2O_3 fraction. The last one improves the possibility for the optical characterization due to the reduction of the absorbance and emission yield without structural properties modification, being determined by the small difference in ionic radii of the rare earth (RE) ions: 0.98 \AA (Nd^{3+}) and 1.03 \AA (La^{3+}) with the coordination number (CN) of $CN_{RE-O} = 8$ [8]. At the same time,

the substitution of Ba^{2+} for Sr^{2+} decreases the crystallisation index of the glasses. Since the values of the radii are 1.42 \AA for Ba^{2+} at $CN_{Ba-O} = 8$ and 1.26 \AA for Sr^{2+} at $CN_{Sr-O} = 6$, the difference in the effective cations ionic radii supports an amorphization. The crystallisation index reduction is in a good agreement with experimental data [9] because the packing density of the $Ba_{1-x}O/Sr_xO-P_2O_5-Al_2O_3-B_2O_3-Nd_2O_3/La_2O_3$ phosphate glasses ($Ba_{1-x}Sr_xgl$) deviates from the linear dependence versus the molar ratio of Sr/Ba. It causes the alkaline earth metal-oxygen polyhedra distortion similarly to that observed in [10], and the transition from the glass net that is built up with BaO_8 polyhedrons to that formed with SrO_6 octahedrons. Such modification effects on the distance between the nearest neighbour La/Nd–La/Nd atomic pairs that can drastically impact on optical properties of the glasses.

In this paper we report the effect of the Ba/Sr ratio on the optical properties of the substituted phosphate laser optical glasses $Ba_{1-x}Sr_xgl$ studied in the whole substitution range $x = 0-1$. It was performed with the molar fraction x increment of 0.25. The glass samples were prepared from the initial reagents purified by specially elaborated chemical techniques. We have applied the nonlinear optical (NLO) response characterization within the spatial beam profile analysis technique [11], which works due to the self-action effect of the continuous wave (CW) and pulsed laser radiation at wavelength 1064 nm (1.17 eV). The approach is based on successful application of the technique for substitution level of the Lu by Gd atoms characterization in the oxyorthosilicates single crystals [12]. It demonstrated a high sensitivity to the oxygen stoichiometry controlled by Sr/Ba content in laser phosphate glasses.

2. Experimental

2.1. Initial materials

Reagent grade Sigma-Aldrich chemicals: $Ba(NO_3)_2$, $Sr(NO_3)_2$, $(NH_4)_2HPO_4$ and

KH_2PO_4 , were purified to prepare a high purity analytical reagent grade chemicals; Onyxmet (Olsztyn, Poland) products: La_2O_3 "D" and Nd_2O_3 "E" was used as is. The following three methods were used for the purification: (i) the complexing of colouring impurities, ppms of *d*-metal ions, Fe, Cu, Co, Ni, Mo, V, Cr and Mn, to form stabile diethyldithiocarbamate (DDC) complexes that adsorbs on non-oxidized activated carbon (AC) derived from wood with further deashed with HF and HCl solution, (ii) the adsorption of the metal ions by the surface of the respective oxidized AC, and (iii) $\text{Ba}(\text{NO}_3)_2$, $\text{Sr}(\text{NO}_3)_2$ reagents recrystallization with the partial precipitation of carbonates that occlude respective *d*-metals ions. $\text{SiO}_2 \cdot n\text{H}_2\text{O}$ was prepared from Dynasylan[®]A, tetraethyl-orthosilicate, that additionally was purified by distillation in the presence of oxidized AC. This purified product was subjected to a hydrolysis in double-distilled water. A small quantity of ammonia (*puriss.*) was added to the water, as the hydrolysis catalyst. The obtained product was separated and then dried. The content of SiO_2 in the product of the hydrolysis was determined by thermogravimetry using an apparatus reported in [13]. The analytical control of the purity was done by the spectrophotometric analysis using an extraction of DDC-complexes with CHCl_3 . As the purification result, the total concentration of colouring impurities in each product is lowered to 0.1–2 ppm from 20–5 ppm for the each of the ions. Boria and alumina used for the glasses preparation are purity reagent grade chemicals of the Federal State Unitary Enterprise IREA (Moscow, Russia).

The obtained purified powders were used for the preparation of glass samples that are based on the composition (in mass%): P_2O_5 – 58, BaO – 12, K_2O – 13, SiO_2 – 2, Al_2O_3 – 8, B_2O_3 – 4, Nd_2O_3 – 0.5, La_2O_3 – 2.5. To prepare the Sr/Ba glasses, the BaO component in the composition of a batch mixture, which is used for the glass preparation, was partially or completely substituted for SrO. A sum of the molar composition of the alkaline earth oxides is kept fixed to the value equivalent to 12 mass% of BaO. The glasses were obtained by the components melting. The prepared glass samples are designated as $\text{Ba}_{1-x}\text{Sr}_x\text{gl}$ and $x = 0, 0.25, 0.75, 0.50$ and 1.0. The value of x denotes the molar fraction of strontium in the glass as regard to the content of Ba.

2.2. Preparation of glasses samples

A typical preparation of glass sample was performed as follows. Appropriate amounts of the initial materials, as per the composition of glasses, were properly weighed. A batch mixture of the initial components (20 g) was wetted with double-distilled water, as mixing medium, mixed in a new porcelain mortar and transferred into a 40 ml transparent quartz crucible. The mixture was heated in a programmable electric resistance furnace, which enclosed volume is made from an alundum tube. The heating was conducted at the slow heating rate up to 1000°C for 4 h. After that, the mixture was heated up to 1330°C for 2 h and periodically mixed with a quartz rod. It should be noticed that the melting temperatures for different compositions were in the range of 1050–1330°C. The melt was maintained at 1330°C in the furnace for 10 min for refining and homogenization and then it was cooled and poured into a cast-iron mould preheated up to 400°C and pressed by a plate. This cast iron mould with a glass within was leaved to get slowly cooled, with a 5°C per minute rate, and after that the resulted glass was annealed in a muffle furnace at the temperature of 450°C for 6 h. The annealing was performed to remove the residual stresses due to temperature gradient, which is produced by rapid cooling. Annealed glass samples were parallel cut to obtain plates of 1 mm thin.

2.3. Samples characterization

Pycnometric density of the glasses (ρ) was determined by weighting, similarly as in ASTM D854 [14], in air and in hexane at 23°C. A non-polar hexane medium was used instead of water to prevent dissolution. The powdered glass samples were mixed with KBr powder and then pressed.

Fourier Transformed Infrared (FTIR) spectroscopy. The pellets were examined by using a Thermo Nicolet Nexus 470 FTIR spectrometer. FTIR spectra were recorded in the wavenumbers range of 400–1600 cm^{-1} at 25°C using a 1026 scans regime.

Ultraviolet-visible (UV-Vis) spectroscopy. UV-VIS spectra of the glasses were recorded by means of UV-VIS spectrophotometer (Shimadzu UV-2700 instrument) by using 1 nm scan step.

Empirical estimation of optical properties of glasses. Optical characteristics of glass materials, in particular, linear and nonlinear refractions, depend on the elec-

tron polarizability of the medium under irradiation [15]. So, the empirical approach proposed by Duffy and Ingram [16] is widely used to bind the chemical composition changes with the optical properties of glasses [16–18]. In this paradigm, the average molar refraction (R_m) was calculated from the equation:

$$R_m = pR_i + qR(O^{2-}) = 2.52 \left(p \sum \alpha_i + q\alpha(O^{2-}) \right), \quad (1)$$

where α_i and $\alpha(O^{2-})$ are the molar polarization ability of cation taken from [8] and anion in a glass matrix, R_i and $R(O^{2-})$ are the ionic constituents of the refraction for cations and oxygen anion, p and q are the cation and anion molar coefficients in the glass formula. The theoretic optical basicity (Λ_{th}), which is sensitive to changes at chemical composition of glass, was estimated through the formula:

$$\Lambda_{th} = \sum \Lambda_i \chi_i, \quad (2)$$

where Λ_i and χ_i are the optical basicity and the molar part of i -component. The polarizability of oxygen ions ($\alpha(O^{2-})$) was calculated through a simple relation with Λ_{th} [19]

$$\alpha(O^{2-}) = (1 - \Lambda_{th}/1.67)^{-1}. \quad (3)$$

Since the linear and nonlinear refractive indexes correlates, the Goldhammer-Herzfeld metallization criterion (GH), which characterizes the covalence of a glass medium, was found from the formula:

$$GH = \left(1 - \frac{R_m}{V_m} \right) = \left(1 - \frac{n_0^2 - 1}{n_0^2 + 2} \right), \quad (4)$$

where V_m is the molar volume, n_0 is the linear refraction index. The ionic packing ratio (V_p) of the glass host- a ratio of specific ion total volume per mole to the molar volume, was calculated, in approximation that each ion in the glass was a rigid sphere, from the equation:

$$V_p = \left(\sum_i \frac{4}{3} \pi r_i^3 n_i N_A \right) V_m^{-1}, \quad (5)$$

where r is the ionic radius, n is the molar fraction, N_A is the Avogadro number, M is the molar weight, and ρ is the pycnometric density of the glass host.

Elastic scattering indicatrix. The optical elastic scattering indicatrices into forward hemisphere of the phosphate glasses were measured under the CW DPSS laser irradiation at 1064 nm. The scattered light was registered by the CCD camera (ATiK 16 IC-HS with pixel size $7.4 \times 7.4 \mu\text{m}^2$ and 640×480 image resolution) at the distance of 39.2 cm from the sample. The elastic scattering losses into forward hemisphere were estimated within the approach described in [20].

Nonlinear optical characterization. The absorptive NLO response was studied by the photoinduced total transmittance variation due to the self-action of the pulsed and CW laser radiation at wavelength 1064 nm (1.17 eV) that can induce resonant two-photon absorption process. The measurements were performed according to the technique described in [11]. For the pulsed excitation we utilized the radiation of the mode-locked Nd³⁺:YAG laser with pulsewidth 42 ps (FWHM) and repetition rate 50 Hz. For CW excitation we applied DPSS laser with peak output power ~ 20 mW. The samples were positioned at the input aperture (1 cm diameter) of the photodiode in order to avoid impact of the scattering extinction losses. We have thoroughly checked the reversibility of the photoinduced transmittance variation with rise/reduction of the input laser radiation intensity.

Fitting the experimental data of the total transmittance versus the peak laser intensity I according to the route described in [21] provides estimation of the photoinduced absorption variation $\Delta\alpha$ magnitudes that is proportional to the imaginary part of the cubic NLO susceptibility $\Delta\alpha \sim \text{Im}(\chi^{(3)})I$.

3. Results and discussion

3.1. Empirical estimation of optical properties

The Ba_{1-x}Sr_xgl pycnometric density, optic and physic characteristics, which are estimated from (1–5), are listed in Table 1. The values of R_m and $\alpha(O^{2-})$ decrease with an increase of the Sr content. GH criterion possesses a significant reduction of the magnitudes up to 0.762 for the laser glasses with $x \geq 0.5$ that show the highest value of a media covalence. The maximal values of ρ and V_m are characterized the pure Ba glass. The value of Λ_{th} decreases monotonously from Ba-based to Sr-based glass.

Table 1. The estimated optical and physical characteristics of the $\text{Ba}_{1-x}\text{Sr}_x\text{gl}$ glasses with different Ba/Sr ratio, Λ_{th} — the theoretic optical basicity, $\alpha(\text{O}^{2-})$ — the polarizability of oxygen ions, R_m — the average molar refraction, V_m — the molar volume, V_p — the ionic packing ratio of the glass host, ρ — the pycnometric density, GH — the Goldhammer-Herzfeld metallization criterion

Sr content, x	0.0	0.25	0.50	0.75	1.0
Λ_{th}	0.639	0.638	0.637	0.636	0.634
$\alpha(\text{O}^{2-})$, cm^3/mol	1.620	1.618	1.616	1.614	1.613
R_m , cm^3/mol	4.788	4.770	4.753	4.736	4.718
V_m , m^3	0.027	0.024	0.020	0.020	0.020
*V_p , %	68.9	59.0	50.1	49.0	49.6
ρ , g/cm^3	3.310	2.930	2.470	2.400	2.410
GH	0.822	0.801	0.762	0.763	0.764

*Note: $\text{CN}_{\text{Ba-O}} = 8$ and $\text{CN}_{\text{Sr-O}} = 6$.

3.2. UV-Vis spectroscopy

The absorption spectra of the studied glasses are presented in Fig. 1 with the most intensive seven Nd^{3+} bands centred at about 805, 744, 684, 583, 525, 428, and 355 nm, which are assigned as in [22]. The absorption bands originate from the ground $^4I_{9/2}$ state due to $4f-4f$ orbital electronic transitions with wave functions localized within the single Nd^{3+} ion. They are forbidden in the electric dipole approximation by quantum mechanical selection rules, which, however, are relaxed via local field-induced intermixing of the f states with higher electronic configurations. The shielding of the $4f$ electrons by the outer complete $5s$ and $5p$ shells results in a quite identical and practically independent position from the minor changes at Nd^{3+} volume concentration with the Ba/Sr ratio.

On the other hand, the substitution of Ba with Sr effects on the structure and properties of the glass matrix. The ionic packing ratio varies slightly for phosphate glasses by changing network modifier ion; therefore, it is difficult to clarify the effect of ionic packing ratio V_p on the optical spectra. The results of studies suggest that Nd^{3+} ions are incorporated into certain sites to break P=O bonds in phosphate glasses.

It should be noted that the glass containing Ba without Sr ($x = 0.0$) is the most transparent among others. The rise of the substitution level up to $x \leq 0.75$ leads to the monotonic darkening of the samples. For the glass containing Sr without Ba ($x = 1.0$) the spectral transmittance of the sample are between the corresponding ones for the glasses with $x = 0.25$ and 0.50 .

We have performed NLO characterization of the samples at the laser radia-

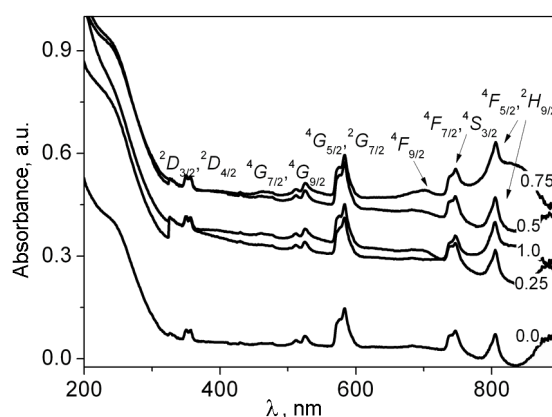


Fig. 1. UV-Vis spectra for the optic $\text{Ba}_{1-x}\text{Sr}_x\text{gl}$ for $x = 0.0, 0.25, 0.50, 0.75, 1.0$ and the assignment of peaks given alongside.

tion wavelength 1064 nm. It can induce resonant two-photon absorption transitions into the band $^4G_{9/2}$ with peak at about 525 nm with efficiency of the photoinduced absorption. The effect allows monitoring the structural modifications of the phosphate glasses produced by the Sr/Ba substitution *via* the laser radiation self-action through the Nd^{3+} resonant excitation.

3.3. FTIR spectroscopy

Normalized IR transmission spectra of the phosphate glass samples with corresponding characteristic vibrations are presented in Fig. 2. Considering the presence of oxygen atoms at the composition of non-phosphate groups (borates, silicates, and aluminates), the ratio of P/O is found to about 3. This ratio corresponds to the composition of metaphosphates, indicating the predominance of tetrahedral Q_2 units as a structural motif of metaphosphate glasses

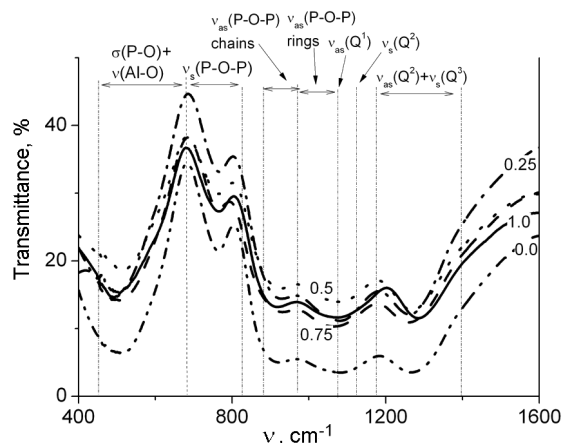


Fig. 2. IR spectra for the optic $\text{Ba}_{1-x}\text{Sr}_x\text{gl}$ glasses with different Sr/Ba content for $x = 0.0, 0.25, 0.50, 0.75, 1.0$.

net [24, 25]. Since Q_2 unit has less amount of the terminal oxygen atoms versus the coordination number of strontium (SrO_6) and barium (BaO_8) ions in oxide glasses, the Sr/Ba ions can coordinate oxygen ions *via* formation of bonds with oxygen ions of corresponding adjacent Q_2 units. Moreover, Sr and Ba atoms can not fill completely the oxygen coordinate environment that indicates the certain oxygen vacancies concentration that suggests to increase with the Ba content rise (x reduction) in the glass matrix. Modification of the optical properties depends on the possibilities of electronic detrapping, which is proportional to the number of related surface and internal oxygen vacancies.

Since the relative intensity of the bands at the replacement of the Sr for Ba and *vice-versa* is changed due to the change of the molar ratios of phosphate units to metal ions, so all the IR spectra are normalized to the molar absorptive coefficient of the media at the same concentration of adsorptive groups. The absorption in the range $430\text{--}1400\text{ cm}^{-1}$ corresponds to vibrations, in the most part, of terminal $Q_1, Q_2,$ and Q_3 groups [24]. So, as can be seen from the Fig. 2, the highest absorption and thus the shortest length of the phosphate chain has the glass with 100 % content of barium ($x = 0.0$). This fact can be explained by a stronger polarizability of the glass matrix that occurs due to the larger ionic radius of the Ba^{2+} ion versus the Sr^{2+} one. The $\text{Ba}_{0.5}\text{Sr}_{0.5}\text{gl}$ demonstrates the lowest absorption that correlates with the metallicity criterion (see Table 1). Due to the long phosphate chains lengths in the phosphate glasses they serve

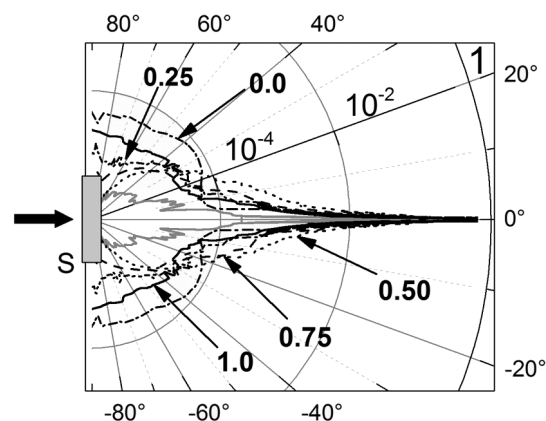


Fig. 3. The optical scattering indicatrices in forward hemisphere at 1064 nm for the optic $\text{Ba}_{1-x}\text{Sr}_x\text{gl}$ glasses with different Sr/Ba content for $x = 0.0, 0.25, 0.50, 0.75, 1.0$. The grey curve corresponds to the freely propagating laser beam as a reference.

as the most asymmetric environment for the Nd^{3+} activator that causes the lowest ionicity of Nd–O bonds. It was confirmed by the results of the UV-Vis absorption studies (see Table 2). For $\text{Ba}_{0.5}\text{Sr}_{0.5}\text{gl}$ a number of oxygen vacancies should be the maximal due to the steric factor [25] that effects on the coordination/existence of long phosphate chains in a net.

3.4. Optical scattering

The optical scattering indicatrices of the studied laser glasses at 1064 nm are represented in Fig. 3. It is shown that the scattering of the glasses strongly depends on the Sr/Ba content. The calculated scattering losses into forward hemisphere and samples thickness L are given in Table 2. It was shown that only the Sr-based and Ba-based glasses are characterized by the smallest scattering losses. The Sr/Ba substitution reveals the nonmonotonic variations of the scattering efficiency: the scattering losses

Table 2. The thicknesses L and the scattering losses into forward hemisphere at 1064 nm for the optic glasses $\text{Sr}_x\text{Ba}_{1-x}\text{gl}$ with different Ba/Sr content

x	$L, \text{ mm}$	Scat. Losses, %
0.0	1.82	2.5
0.25	1.94	8.4
0.50	2.00	12.6
0.75	1.82	8.4
1.0	1.76	5.2

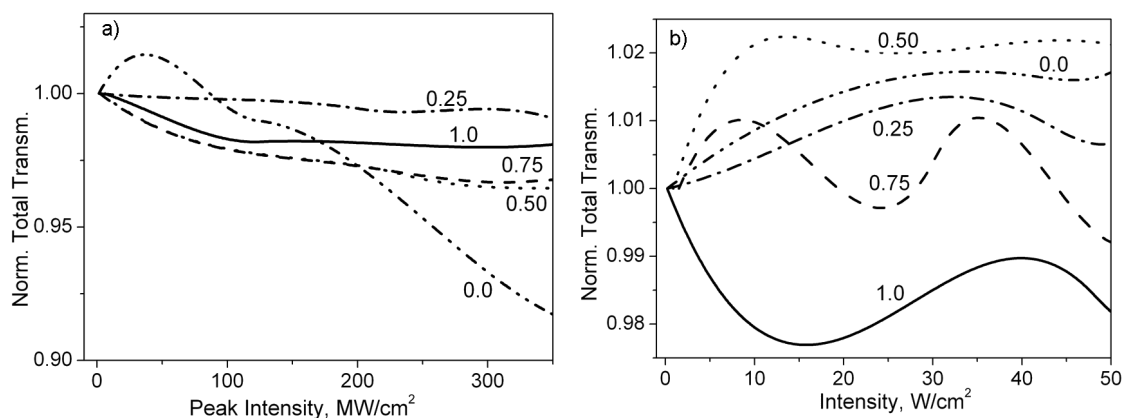


Fig. 4. The photoinduced variations of the normalized total transmittance in the pulsed (42 ps) — (a), and CW — (b) excitation regimes at 1064 nm for the optic $Ba_{1-x}Sr_x$ gl glasses with different Sr/Ba content for $x = 0.0, 0.25, 0.50, 0.75,$ and 1.0 .

are the highest for the $x = 0.5$ (~ 12.6 %) and decreases for the other glasses. For the Sr/Ba content $x = 0.25$ and $x = 0.75$ the scattering losses is almost equal.

3.5. Nonlinear optical characterization

The photoinduced variations of the total transmittance of the phosphate laser glasses within the pulsed excitation regime are presented in Fig. 4a. Each experimental curve corresponds to the data acquisition for about 5000 laser shots. The typical error bar of the curves is about ± 1 %. It can be seen that the total transmittance of the glass with 100 % of Ba significantly differs versus the others. It demonstrates the photoinduced bleaching at the peak intensity range 5–50 MW/cm^2 . For the higher peak intensities the process saturates and turns to efficient photodarkening effect with magnitude ~10 %. For the samples that contain Sr the monotonic photoinduced darkening effect was observed with the saturation at about 100 MW/cm^2 . The efficiency of the process significantly depends on the Sr/Ba ratio. The almost total compensation of the pronounced photoinduced absorption variations was revealed for the $Ba_{0.75}Sr_{0.25}gl$ sample.

For the case of the CW laser excitation the photoinduced variations of the total transmittance are presented in Fig. 4b. It was shown that only for the degenerate case $x = 1$ the glass demonstrates the photodarkening effect that turns to photobleaching one at about 15 W/cm^2 . For the samples that contain Ba the photoinduced bleaching was observed. The process saturates and turns to photodarkening at the following intensity magnitudes that rises with

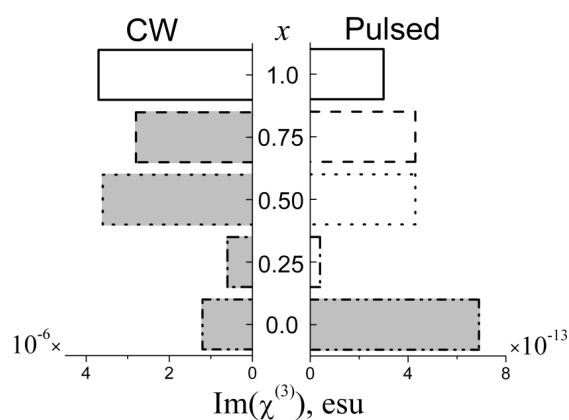


Fig. 5. The magnitudes of the imaginary part of the cubic NLO susceptibility in CW and pulsed excitation regimes for the optic $Ba_{1-x}Sr_x$ gl glasses with $x = 0.0, 0.25, 0.50, 0.75,$ and 1.0 . The transparent bars indicate the photoinduced darkening effect ($\Delta\alpha > 0$), patterned — the photoinduced bleaching one ($\Delta\alpha < 0$).

the reduction of parameter x given in brackets: ~8.0 W/cm^2 (0.75); ~13.3 W/cm^2 (0.5); ~32.4 W/cm^2 (0.25), and ~34.3 W/cm^2 (0.0).

From the obtained total transmittance dependences the magnitudes of the imaginary part of the cubic NLO susceptibility $Im(\chi^{(3)})$ were estimated. The comparison of the results for the cases of the pulsed and CW laser excitations are represented in Fig. 5 as a bar diagram. The transparent bars indicate the photoinduced darkening effect ($\Delta\alpha > 0$), patterned — the photoinduced bleaching one ($\Delta\alpha < 0$).

It was shown that for CW excitation the substitution of Sr for Ba results in $Im(\chi^{(3)})$

reduction and the sign turn. In the case of the pulsed excitation the increase of the Ba content reveals the $\text{Im}(\chi^{(3)})$ rise except the glass with $x = 0.25$.

The lowest magnitudes of the photoinduced bleaching/absorption efficiencies $|\text{Im}(\chi^{(3)})| \sim 6 \cdot 10^{-7}/4 \cdot 10^{-14}$ esu for CW/pulsed excitation regime was revealed for $\text{Ba}_{0.75}\text{Sr}_{0.25}\text{g}$ that indicates it to be the most optimal against the laser beam distortion inside the media.

The observed sensitivity of the laser radiation self-action effects to the Sr/Ba ratio in the phosphate glass indicates the applied approach to be promising for their structure diagnostics. The proposed technique can be applied for the oxygen stoichiometry diagnostics for laser phosphate glasses due to the resonant multiphoton excitation of Nd^{3+} ions at 1064 nm.

4. Conclusions

The characterization of the optic phosphate glasses $\text{Ba}_{1-x}\text{Sr}_x\text{g}$ with different content of Sr/Ba ions was performed by UV-VIS spectral, FTIR, elastic scattering indicatrices and NLO response analyses. Advanced methods for purification of initial reagents were applied. The effect of the Sr/Ba ratio on the glass structure was observed. It was shown that the positions and the form of UV-Vis spectra peaks of the studied optic glasses are similar and almost independent on the minor Nd^{3+} volume concentration variations with the Ba/Sr ratio.

The optical scattering indicatrices analysis revealed the nonmonotonic variations of the scattering efficiency versus the Sr/Ba ratio: the scattering losses $\sim 12.6\%$ are the highest for the $\text{Ba}_{0.5}\text{Sr}_{0.5}\text{g}$ that corresponds to the peak oxygen vacancies concentration due to the steric factor. The glasses with symmetric substitution levels $x = 0.25$ and 0.75 demonstrate the similar level of extinction losses $\sim 8.4\%$.

The absorptive NLO response under the excitation of CW and pulsed laser radiation at 1064 nm were analyzed. It was shown that the photoinduced absorption/bleaching variations are strongly dependent on the ratio of Sr/Ba in the phosphate glasses. The observed effects were attributed to the resonant two-photon absorption into Nd^{3+} band ${}^4G_{9/2}$. The efficiency of the transition can be controlled by the local field deterioration due to the oxygen vacancies impact in the vicinity of the Nd ion. Except the degenerate cases at $x = 0.0$ and 1.0 the $\text{Ba}_{0.5}\text{Sr}_{0.5}\text{g}$

reveals the most efficient photoinduced absorption/bleaching efficiency within pulsed/CW laser excitation at 1064 nm correspondingly. In degenerate cases for both excitation regimes we have observed photobleaching ($x = 0$, $\text{Im}(\chi^{(3)}) < 0$) and photoinduced absorption ($x = 1$, $\text{Im}(\chi^{(3)}) < 0$) at the initial range of the laser intensities. For the rest of the samples the absorptive NLO response has opposite signs within laser pulsed and CW excitation (see Fig. 5).

The lowest magnitudes of the photoinduced bleaching/absorption efficiencies $|\text{Im}(\chi^{(3)})| \sim 6 \cdot 10^{-7}/4 \cdot 10^{-14}$ esu for CW/pulsed excitation regime was revealed for $\text{Ba}_{0.75}\text{Sr}_{0.25}\text{g}$ that indicates it to be the most optimal against the laser beam distortion inside the media. For the $x = 0.25$ composition it was also obtained UV-vis range absorption and elastic scattering reduction. The mentioned optical properties improvement was achieved by both the substitution level choice modification and the purification of the initial reagents.

On the basis of the obtained results we suggest a high potential of the application of the CW and pulsed laser radiation self-action effects at wavelength 1064 nm within elastic scattering indicatrices analysis for the testing of oxygen vacancies concentration in the phosphate laser glasses for the application in the field of quantum and optoelectronics.

Acknowledgments. The work was partially supported by joint NASU-NASB project No.04-02-15, and NASU project B/166.

References

1. E.U. Patent 0,218,135 (1987).
2. U.S. Patent 3,979,322 (1976).
3. R.F. Patent 2,426,701 (2011).
4. U.S. Patent 5526369A (1996).
5. U.S. Patent 20,060,128,549 (2008).
6. U.S. Patent 7,435,695B2 (2008).
7. U.S. Patent 4,075,120A (1978).
8. R.D.Shannon, R.X.Fischer, *Phys. Rev. B*, **73**, 235111 (2006).
9. G.Walter, U.Hoppe, A.Barz et al., *J.Non-Cryst. Solids.*, **263–264**, 48 (2000).
10. A.Paleari, V.N.Sigaev, N.V.Golubev et al., *Mater. Chem. Phys.*, **128**, 12 (2011).
11. V.Ya.Gayvoronsky, L.A.Golovan, M.A.Kopylovsky et al., *Quant. Electron.* **41**, 257 (2011).
12. A.V.Uklein, A.S.Popov, V.V.Multian et al., *NanoScale Res. Lett.*, **10**, 102 (2015).
13. V.E.Diyuk, A.N.Zaderko, K.I.Veselovska, V.V.Lisnyak, *J. Thermal Anal. Calorim.*, **120**, 1665 (2015).
14. ASTM D854-14 (2014).

15. I.V.Dimitrov, S.Sakka, *J. Appl. Phys.*, **79**, 1736 (1996).
16. J.A.Duffy, M.D.Ingram, *J. Am. Chem. Soc.*, **93**, 6448 (1971).
17. L.S.Dent-Glasser, J.A.Duffy, *J. Chem. Soc., Dalton Trans.*, **10**, 2323 (1987).
18. J.A.Duffy, *J. Phys. Chem.*, **110**, 13245 (2006).
19. Y.Waseda, J.M.Toguri, *The Structure and Properties of Oxide Melts: Application of Basic Science to Metallurgical Processing*, World Scientific Publishers, Singapore (1998).
20. V.Ya.Gayvoronsky, A.S.Popov, M.S.Brodyn et al., in: *Nanocomposites, Nanophotonics, Nanobiotechnology and Applications*, Edition: Springer Proceedings in Physics 156, Springer, Cham-Heidelberg-NY (2015), p.147.
21. V.Ya.Gayvoronsky, M.A.Kopylovsky, M.S.Brodyn et al., in: *Nanomaterials Imaging Techniques, Surface Studies and Applications*, Edition: Springer Proceedings in Physics 146, Springer, New York, USA (2013), p.349.
22. A.B.P.Lever, *Inorganic Electronic Spectroscopy. Studies in Physical and Theoretical Chemistry*, Elsevier, Amsterdam (1986).
23. J.J.Garcia Sole, L.E.Bausa, D.Jaque, *An Introduction to the Optical Spectroscopy of Inorganic Solids*, Wiley, Chichester (2005).
24. L.L.Velli, C.P.E.Varsamis, E.I.Kamitsos et al., *Phys. Chem. Glasses*, **46**, 178 (2005).
25. H.Takebe, Y.Nageno, K.Morinaga, *J. Am. Ceram. Soc.* **77**, 2132 (1994).