The demixing phase instability in the ion-dipole mixture

X.S.Chen¹, F.Forstmann²

- Institute of Theoretical Physics, Academia Sinica, P.O. Box 2735, Beijing 100080, China
- Institut für Theoretische Physik, Freie Universität Berlin, Arnimallee 14, D-14195 Berlin, Germany

Received November 2, 2001

By considering the variation of the grand potential functional $\,\Omega\,$ with respect to small density fluctuations $\,\delta\rho_{\alpha}(1)\,$ we can determine the phase instability of a system from the correlation functions. Using the reference hypernetted chain (RHNC) approximation the correlation functions of an ion-dipole mixture have been calculated. With the obtained correlation functions the phase behaviour of ion-dipole mixtures has been investigated.

Key words: phase transition, demixing, instability, ion-dipole mixture, integral equations

PACS: 61.20.Gy, 64.60.-i, 64.75.+g, 64.70.-p, 64.30.+t, 61.20.Q

This article is dedicated to our dear friend Jean-Pierre Badiali on the occasion of his 60th birthday.

1. Introduction

A mixture of charged hard spheres and dipolar hard spheres is a simple model of an electrolyte. For such a model a series of statistical mechanical investigations have been done with integral equations as well as with simulations. In the mean spherical approximation (MSA) the analytic solution of the ion-dipole mixture can be obtained [1,2]. Using the MSA results the phase transition in ion-dipole mixture has been discussed by Høye et al. [3] and Harvey [4]. But it was known that MSA results are not sufficient for the ion-dipole mixture because of the strength of the interactions. The correlation functions of ion-dipole mixtures can also be calculated in the so-called linearized, quadratic [5] and reference hypernetted-chain [6] (LHNC, QHNC and RHNC) approximations with numerical iteration methods. Comparing with the Monte Carlo simulation results [7–9] only the RHNC approximation works very well [6,7], and here we will calculate the correlation functions in the RHNC approximation.

In the RHNC approximation, for moderate dipolar moments a convergent solution for ion-dipole mixtures cannot be obtained when the charge of the ions is still smaller than an elementary charge. This was already recognized by Dong et al. [10]. Also the simulations encountered problems in this range of parameters [11].

By analyzing the phase stability of the system with respect to the small density fluctuations of ions and dipoles we have shown that this problem is related to the phase instability of the ion-dipole mixture [12]. In this article we will give a more detailed investigation of the ion-dipole phase instability.

The article is organized as follows. In section 2 we define the ion-dipole mixture model in detail. Section 3 discusses the calculation of the correlation functions. In section 4 we describe the method for the phase instability analysis. We use the calculated correlation functions to discuss the phase instability of the ion-dipole mixture in section 5. Section 6 contains the conclusions. The appendix describes the treatment of the long range tails of the interactions and the correlations.

2. The model

Our model electrolyte is a mixture of hard sphere ions and hard sphere dipoles. The ions have the charge $q_i(q_+ = q, q_- = -q)$ and number density ρ_i . The dipoles have dipole moment $\vec{\mu}$ and number density ρ_d . Both ions and dipoles have the same hard sphere diameter σ . The system is taken to be electrically neutral:

$$\sum_{i} q_i \rho_i = 0. \tag{2.1}$$

The pair potential between two particles in the system can be written as

$$u_{ij}(12) = u^{HS}(12) + q_i q_j / r,$$
 (2.2)

$$u_{id}(12) = u^{HS}(12) - (q_i \mu/r^2)(\hat{\mu}_2 \cdot \hat{r}_{12}),$$
 (2.3)

$$u_{dd}(12) = u^{HS}(12) - (\mu^2/r^3)D(12).$$
 (2.4)

The variables 1, 2 imply position and orientation (for the dipoles) with $2 \equiv (\vec{r}_2, \hat{\mu}_2)$, where $\hat{\mu}_2$ is the unit vector in the direction of the dipole moment at \vec{r}_2 with the Euler angles $\omega_2 \equiv (\theta_2, \varphi_2)$. $r = |\vec{r}_2 - \vec{r}_1|$ and \hat{r}_{12} is the unit vector pointing from position 1 to position 2: $\hat{r}_{12} = (\vec{r}_2 - \vec{r}_1)/r$, with its Euler angles denoted as ω_r . $u^{\text{HS}}(12)$ is the hard sphere interaction defined by

$$u^{\text{HS}}(12) = \begin{cases} \infty, & r < \sigma \\ 0, & r > \sigma. \end{cases}$$
 (2.5)

The function D(12) is the angle dependent part of dipole-dipole interaction and is defined as

$$D(12) = 3(\hat{\mu}_1 \cdot \hat{r}_{12})(\hat{\mu}_2 \cdot \hat{r}_{12}) - \hat{\mu}_1 \cdot \hat{\mu}_2.$$
 (2.6)

3. The calculation of the correlation functions

The correlation functions of ion-dipole mixtures can be calculated with reliable results by the Ornstein-Zernike (O.Z.) equation and the reference hypernetted chain (RHNC) closure. This conclusion was reached by Caillol et al. [7]. They have done Monte Carlo simulations for the ion-dipole mixture and have shown that the RHNC approximation works well.

The Ornstein-Zernike equation gives the relation between the total correlation functions $h_{\alpha\beta}(12)$ and the direct correlation functions $c_{\alpha\beta}(12)$

$$h_{\alpha\beta}(12) = c_{\alpha\beta}(12) + \sum_{\gamma} \int d3 \, h_{\alpha\gamma}(13) \rho_{\gamma}(3) c_{\gamma\beta}(32) \,.$$
 (3.1)

Here $\rho_{\alpha}(1)$ is the number of the particles α at the position \vec{r}_1 pointing in the direction ω_1 per unit volume and unit space angle. The number density $\rho_{\alpha}(\vec{r}_1)$ is obtained by integrating $\rho_{\alpha}(1)$ over all orientations ω_1 . For the homogeneous isotropic system we write simply ρ_{α} for the number density and $\rho_{\alpha}(1) = \rho_{\alpha}/4\pi$. For later angular expansions (section 4) it is convenient to extend these definitions also to the ion densities $\rho_{+}(1)$ and $\rho_{-}(1)$. We then have everywhere $\rho_{\pm}(1) = \rho_{\pm}(\vec{r}_1)/4\pi$.

Because the O.Z. equation contains two unknown functions, another relation between direct and total correlation functions is needed to calculate the correlation functions. This relation can be written as

$$1 + h_{\alpha\beta}(12) = \exp[-u_{\alpha\beta}(12)/k_{\rm B}T + h_{\alpha\beta}(12) - c_{\alpha\beta}(12) + b_{\alpha\beta}(12)], \tag{3.2}$$

where $b_{\alpha\beta}(12)$ is the so-called bridge term. Usually the bridge term cannot be known exactly and an approximation must be made. If we take $b_{\alpha\beta}(12) = 0$ the hypernetted chain (HNC) approximation is obtained. In the reference hypernetted chain approximation (RHNC) [13], we approximate $b_{\alpha\beta}$ in equation (3.2) by the bridge term of a reference system. Here we choose the hard sphere fluid as our reference system, which is obtained by switching off the Coulomb and dipole interactions.

The bridge term of the reference system $b_{\alpha\beta}^{\rm R}(12)$ can be expressed as

$$b_{\alpha\beta}^{\rm R}(12) = \ln[1 + h^{\rm HS}(12)] - h^{\rm HS}(12) + c^{\rm HS}(12), \qquad r > \sigma,$$
 (3.3)

where $h^{\rm HS}(12)$ and $c^{\rm HS}(12)$ are the total and direct correlation function of hard spheres and can be calculated using Verlet and Weis's method [14].

Because of the angle dependence of the ion-dipole and dipole-dipole interactions the correlation functions of the ion-dipole and dipole-dipole relations are also angle dependent. To treat such correlation functions we will expand them by the so-called spherical invariants $\Phi^{l_1 l_2 l}(\omega_1 \omega_2 \omega_r)$, which are linear combinations of spherical harmonics [15–17]. We follow the notation of Gray and Gubbins [16]

$$f_{\alpha\beta}(12) = \sum_{l_1 l_2 l} f_{\alpha\beta}^{l_1 l_2 l}(r_{12}) \Phi^{l_1 l_2 l}(\omega_1 \omega_2 \omega_r),$$
 (3.4)

$$\Phi^{l_1 l_2 l_r}(\omega_1 \omega_2 \omega_r) = \sum_{\mu \nu \lambda} C(l_1 l_2 l; \mu \nu \lambda) D^{l_1}_{\mu 0}(\omega_1) D^{l_2}_{\nu 0}(\omega_2) Y_{l \lambda}(\omega_r), \tag{3.5}$$

where f = h or c and $C(l_1 l_2 l_r; \mu \nu \lambda)$ is a Clebsch-Gordan coefficient.

In some situations it is convenient to use the " \vec{r} -frame", which corresponds to choosing the polar axis along the intermolecular axis [16]. In \vec{r} -frame the correlation functions are expanded as

$$f_{\alpha\beta}(12) = \sum_{l_1 l_2 m} f_{\alpha\beta}(l_1 l_2 m, r) D_{m0}^{l_1}(\omega_1') D_{\underline{m}0}^{l_2}(\omega_2'), \tag{3.6}$$

 ω_i' are Euler angles of the dipoles relative to \vec{r}_{12} as polar axis. The \vec{r} -frame coefficients $f_{\alpha\beta}(l_1l_2m,r)$ are related to the space-fixed frame coefficients $f_{\alpha\beta}^{l_1l_2l}(r)$ via

$$f_{\alpha\beta}(l_1 l_2 m, r) = \sum_{l} \left(\frac{2l+1}{4\pi}\right)^{\frac{1}{2}} C(l_1 l_2 l; m\underline{m}0) f_{\alpha\beta}^{l_1 l_2 l}(r) . \tag{3.7}$$

In the isotropic bulk problem, where the particle density $\rho_{\alpha}(1)$ doesn't depend on position and orientation, the Ornstein-Zernike equation is used usually after Fourier transformation. After spherical harmonic expansion and Fourier transform the Ornstein-Zernike equation can be written in a matrix form

$$\mathbf{H}(m,k) = \mathbf{C}(m,k) + \mathbf{H}(m,k)\mathbf{C}(m,k), \tag{3.8}$$

where the matrices \mathbf{F} ($\mathbf{F} = \mathbf{H}, \mathbf{C}$) are defined by the \vec{k} -frame coefficients [15–17] of correlation functions, $\tilde{f}_{\alpha\beta}(l_1l_2m, k)$ (f = h, c) with

$$F(m,k)_{ij} = (-)^m (2l_1 + 1)^{\frac{1}{2}} (2l_2 + 1)^{\frac{1}{2}} \rho_{\alpha}^{\frac{1}{2}} \rho_{\beta}^{\frac{1}{2}} \tilde{f}_{\alpha\beta}(l_1 l_2 m, k). \tag{3.9}$$

Here the indices $i = (\alpha, l_1), j = (\beta, l_2)$ are combinations of particle species index and angular index. The \vec{k} -frame coefficients $\tilde{f}_{\alpha\beta}(l_1l_2m, k)$ are calculated in the following way from the \vec{r} -frame coefficients

$$\tilde{f}_{\alpha\beta}(l_1 l_2 m, k) = \sum_{l} \left(\frac{2l+1}{4\pi}\right)^{\frac{1}{2}} C(l_1 l_2 l; m\underline{m}0) \tilde{f}_{\alpha\beta}^{l_1 l_2 l}(k), \qquad (3.10)$$

$$\tilde{f}_{\alpha\beta}^{l_1 l_2 l}(k) = 4\pi i^l \int_0^\infty r^2 dr j_l(kr) f_{\alpha\beta}^{l_1 l_2 l}(r), \qquad (3.11)$$

where $j_l(kr)$ are the spherical Bessel functions.

In the closure equation (3.2) there are angular dependences in the exponent, therefore we will have difficulty when using the expansion by spherical harmonics. To overcome this difficulty Fries and Patey [6] proposed to differentiate the closure with respect to the particle distance $r = |\vec{r}_2 - \vec{r}_1|$. Writing equation (3.2) as $1 + h_{\alpha\beta} = \exp(h_{\alpha\beta} - c'_{\alpha\beta}) = \exp[\eta'_{\alpha\beta}]$ and differentiating it we get

$$\frac{\partial c'_{\alpha\beta}(12)}{\partial r} = h_{\alpha\beta}(12) \frac{\partial \eta'_{\alpha\beta}(12)}{\partial r}$$
 (3.12)

with functions $c'_{\alpha\beta}(12)$ and $\eta'_{\alpha\beta}(12)$ defined as

$$c'_{\alpha\beta}(12) = c_{\alpha\beta}(12) + u_{\alpha\beta}(12)/k_{\rm B}T - b_{\alpha\beta}(12),$$
 (3.13)

$$\eta'_{\alpha\beta}(12) = h_{\alpha\beta}(12) - c'_{\alpha\beta}(12) .$$
 (3.14)

Here we derive the equations in general. The RHNC approximation will be obtained when setting $b_{\alpha\beta}(12) = b_{\alpha\beta}^{\rm R}(12)$ according to equation (3.3). After harmonic expansion the closure can be expressed with \vec{r} -frame coefficients as

$$\frac{\partial c'_{\alpha\beta}(l_1 l_2 m, r)}{\partial r} = \sum_{l_3 l_4 m_1, l_5 l_6 m_2} F^{l_1 l_2 m}_{l_3 l_4 m_1, l_5 l_6 m_2} h_{\alpha\beta}(l_3 l_4 m_1, r) \frac{\partial \eta'_{\alpha\beta}(l_5 l_6 m_2, r)}{\partial r}, \quad (3.15)$$

where the factors $F_{l_3l_4m_1,l_5l_6m_2}^{l_1l_2m}$ are the products of Clebsch-Gordan coefficients

$$F_{l_3 l_4 m_1, l_5 l_6 m_2}^{l_1 l_2 m} = C(l_3 l_5 l_1; m_1 m_2 m) C(l_4 l_6 l_2; m_1 m_2 m) \times C(l_3 l_5 l_1; 000) C(l_4 l_6 l_2; 000).$$
(3.16)

Later Calliol [18] proposed to differentiate the closure with respect to the orientation of the particles. For example we can differentiate the closure with respect to the orientation of particle 2 by the angular momentum operator

$$J_{+}(2) = L_{x}(2) + iL_{y}(2) \tag{3.17}$$

and obtain

$$J_{+}(2)c'_{\alpha\beta}(12) = h_{\alpha\beta}(12)J_{+}(2)\eta'_{\alpha\beta}(12). \tag{3.18}$$

Using the \vec{r} -frame coefficients the closure can be expressed as

$$c'_{\alpha\beta}(l_1 l_2 m, r) = \sum_{l_3 l_4 m_1, l_5 l_6 m_2} G^{l_1 l_2 m}_{l_3 l_4 m_1, l_5 l_6 m_2} h_{\alpha\beta}(l_3 l_4 m_1, r) \eta'_{\alpha\beta}(l_5 l_6 m_2, r)$$
(3.19)

with the coefficients $G_{l_3l_4m_1,l_5l_6m_2}^{l_1l_2m}$ defined as

$$G_{l_3 l_4 m_1, l_5 l_6 m_2}^{l_1 l_2 m} = C(l_3 l_5 l_1; m_1 m_2 m) C(l_4 l_6 l_2; m_1 m_2 m) C(l_3 l_5 l_1; 000)$$

$$\times C(l_4 l_6 l_2; 011) [l_6 (l_6 + 1)]^{\frac{1}{2}} [l_2 (l_2 + 1)]^{-\frac{1}{2}}.$$
(3.20)

The advantage of Caillol's method is that we get a set of equations directly for the expansion coefficients and not for their derivatives as in the procedure introduced by Fries and Patey. But equations (3.18) and (3.19) fail to calculate the coefficient belonging to $l_1 = l_2 = 0$. In our calculation here we use the combination of the two methods. For the $l_1 = l_2 = 0$ term we use Fries's method and for the other terms we use Caillol's method. Combining equations (3.8), (3.15) and (3.19) we can calculate the correlation functions by iteration. We start with a guess for c, evaluate h from (3.8) and get an improved c from (3.15) and (3.19).

Details of the treatment of the long range tails of interactions and correlation functions can be found in the Appendix.

4. The method for analysis of the instability

The grand potential functional of an ion-dipole mixture can be written as

$$\Omega(T, V, \{\mu_{\alpha}\}; \{\rho_{\alpha}\}) = F_{\mathrm{id}}[\rho] + F_{\mathrm{ex}}[\rho] + \int d1 \sum_{\alpha} W_{\alpha}(1)\rho_{\alpha}(1), \tag{4.1}$$

where $F_{\rm id}[\rho]$ is the Helmholtz free energy functional of the ideal gas

$$F_{\rm id}[\rho] = k_{\rm B}T \int d1 \sum_{\alpha} \rho_{\alpha}(1) \left[\ln \left(4\pi \Lambda_{\alpha}^{3} \rho_{\alpha}(1) \right) - 1 \right]$$
 (4.2)

with $\Lambda_{\alpha} = h/(2\pi m_{\alpha}k_{\rm B}T)^{1/2}$ the thermal de Broglie wavelength. $F_{\rm ex}[\rho]$ is the excess Helmholtz free energy functional, which represents the contribution from the interactions between the particles. The sum over α in equation (4.1) is done for plus ion, minus ion and dipole. $W_{\alpha}(1) = \widetilde{W}_{\alpha}(1) - \mu_{\alpha}$ with chemical potential μ_{α} and external potential $\widetilde{W}_{\alpha}(1)$. The equilibrium density $\rho_{\alpha}(1)$ is determined by the minimum of the grand potential functional for fixed T, V, μ_{α} [19,20].

To study the stability of the equilibrium state we consider the fluctuation of the grand potential around the minimum caused by small density fluctuations $\delta \rho_{\alpha}(1)$ [17,21]. We use the functional Taylor expansion to the second order

$$\delta\Omega = \Omega[T, V, \mu_{\alpha}; \rho_{\alpha} + \delta\rho_{\alpha}] - \Omega[T, V, \mu_{\alpha}; \rho_{\alpha}]
\approx \frac{1}{2} \int \int \sum_{\alpha\beta} \frac{\delta^{2}\Omega}{\delta\rho_{\alpha}(1)\delta\rho_{\beta}(2)} \Big|_{\text{Equilib}} \delta\rho_{\alpha}(1)\delta\rho_{\beta}(2) d1d2 .$$
(4.3)

The second functional derivatives of Ω are related to the direct correlation functions [22]

$$\frac{\delta^2 \Omega}{\delta \rho_{\alpha}(1)\delta \rho_{\beta}(2)} \bigg|_{\text{Equilib}} = k_{\text{B}} T \left[\frac{\delta_{\alpha\beta}\delta(1,2)}{\rho_{\alpha}(1)} - c_{\alpha\beta}(1,2) \right]. \tag{4.4}$$

In order to treat also angular dependent correlation functions, as we need for dipolar interactions, we expand the particle density fluctuation $\delta \rho_{\alpha}(1)$ as well as the direct correlation functions $c_{\alpha\beta}(1,2)$ in spherical harmonics

$$\delta \rho_{\alpha}(1) = \sum_{lm} \delta \rho_{\alpha}^{lm}(\vec{r}_1) Y_{lm}(\omega_1), \qquad (4.5)$$

$$c_{\alpha\beta}(1,2) = \sum_{l_1l_2l} c_{\alpha\beta}^{l_1l_2l}(r_{12}) \Phi^{l_1l_2l}(\omega_1\omega_2\omega_r) . \tag{4.6}$$

The integral in equation (4.3) is a convolution, so we get a product of the corresponding Fourier components after the Fourier transform. Using the expansions (4.5) and (4.6) in equation (4.3) and finishing the angle integration over ω_1 and ω_2 we obtain

$$\delta\Omega = \frac{k_{\rm B}T}{2} \frac{1}{(2\pi)^3} \int d^3k \sum_{m} \sum_{\alpha\beta} \sum_{l_1 l_2} 4\pi \delta \tilde{\rho}_{\alpha}^{l_1 m}(k) \rho_{\alpha}^{-\frac{1}{2}} \Big[\delta_{\alpha\beta} \delta_{l_1 l_2} - (-1)^m \tilde{c}_{\alpha\beta}^{l_1 l_2 m}(k) \rho_{\beta}^{\frac{1}{2}} \rho_{\beta}^{\frac{1}{2}} (2l_1 + 1)^{-\frac{1}{2}} (2l_2 + 1)^{-\frac{1}{2}} \Big] \rho_{\beta}^{-\frac{1}{2}} \delta \tilde{\rho}_{\beta}^{l_2 m}(k)^* . \quad (4.7)$$

Equation (4.7) is a quadratic form of the fluctuations vector $\delta\rho(k)$ with $\delta\rho_{\mu}(k) = \delta\tilde{\rho}_{\alpha}^{l_1m}(k)\rho_{\alpha}^{-\frac{1}{2}}(4\pi)^{\frac{1}{2}}$. The index $\mu = (\alpha, l_1m)$ is a combination of particle species index and angular indices. The coefficients of the quadratic form yield the matrix $\mathbf{M}(k)$. The k-integral can be replaced by a sum over discrete k-values if we imagine periodic boundary conditions: $\frac{1}{(2\pi)^3}\int \mathrm{d}^3k \to \frac{1}{V}\sum_k$. Therefore we have

$$\delta\Omega = \frac{k_{\rm B}T}{2} \frac{1}{V} \sum_{k} \delta\rho(k) \mathbf{M}(k) \delta\rho^{\dagger}(k). \tag{4.8}$$

In the case of angle dependent interactions, $\mathbf{M}(k)$ is diagonal with respect to the angular index m and the matrix has a block form. Therefore equation (4.8) can be decomposed into a sum over submatrix products for separate m

$$\delta\Omega = \frac{k_{\rm B}T}{2} \frac{1}{V} \sum_{k} \sum_{m} \delta\rho(m, k) \mathbf{M}(m, k) \delta\rho^{\dagger}(m, k) = \sum_{k} \sum_{m} \delta\Omega(m, k) . \tag{4.9}$$

Now the index of vector and matrix is only a combination of particle species index and angular index l_1 . The matrices $\mathbf{M}(m,k)$ are Hermitian and can be expressed by the direct correlation function matrices $\mathbf{M}(m,k) = \mathbf{I} - \mathbf{C}(m,k)$ according to equation (3.9). They become real, if in equation (4.6) there exist only terms with even l-values. This is the case for pure dipolar systems or for mixtures of dipoles and neutral particles, but not for mixtures of dipoles and ions.

The probability distribution $P(\delta \rho(m,k))$ for fluctuations is Gaussian

$$P(\delta\rho(m,k)) \sim \exp\left(-\frac{\delta\Omega(m,k)}{k_{\rm B}T}\right) = \exp\left(-\frac{1}{2V}\delta\rho(m,k)\mathbf{M}(m,k)\delta\rho^{\dagger}(m,k)\right) \tag{4.10}$$

and the mean values of the fluctuation products are

$$\langle \delta \rho_{\mu}(m,k) \delta \rho_{\nu}^{\dagger}(m,k) \rangle = V(\mathbf{M}^{-1})_{\mu\nu}(m,k) = V[\mathbf{I} - \mathbf{C}(m,k)]_{\mu\nu}^{-1}. \tag{4.11}$$

Since fluctuation $\delta\rho(m,0)$ is proportional to V and \mathbf{C} is independent of V, the homogeneous density fluctuations $\langle\delta\rho_{\alpha}^2\rangle^{\frac{1}{2}}$ decay like $V^{-\frac{1}{2}}$, as they should [23]. Using the Ornstein-Zernike equation (3.8) the mean values of the fluctuation products can also be expressed by the total correlation function matrices

$$\langle \delta \rho(m,k) \delta \rho^{\dagger}(m,k) \rangle = V[\mathbf{I} + \mathbf{H}(m,k)].$$
 (4.12)

For complete insight into the phase instability of fluid mixtures we must diagonalize the matrix $\mathbf{M}(m,k)$ in equation (4.9). With the eigenvalues $\lambda_{\sigma}(m,k)$ and the normalized eigenvectors $\vec{x}_{\sigma}(m,k)$ of $\mathbf{M}(m,k)$ we can write

$$\delta\Omega = \frac{k_{\rm B}T}{2} \frac{1}{V} \sum_{k} \sum_{m} \sum_{\sigma} \lambda_{\sigma}(m,k) |\delta\rho_{\sigma}'(m,k)|^{2}, \qquad (4.13)$$

$$\delta \rho_{\sigma}'(m,k) = \sum_{\mu} x_{\sigma,\mu}^*(m,k) \delta \rho_{\mu}(m,k) . \qquad (4.14)$$

For stable phases all eigenvalues are positive and any fluctuation of a one-particle density will increase the grand potential Ω . If one eigenvalue $\lambda_{\sigma}(k)$ approaches zero, then there is a fluctuation $\delta \rho'_{\sigma}(m,k)$ which doesn't increase Ω . In this case the system becomes unstable and the mean value of this $\delta \rho'_{\sigma}(mk)$ becomes infinite.

For a pure gas-liquid phase transition the most unstable fluctuation $\delta\rho_{\sigma}(m,k)$ should be the fluctuation of the total density and for a purely demixing phase transition the most unstable fluctuation should be a concentration fluctuation. In general the most unstable fluctuation $\delta\rho'_{\sigma}(m,k)$ (equation (4.14)) belonging to the smallest eigenvector λ_{σ} , which eventually goes to zero, is a linear combination of all kinds of density fluctuations. The phase instability is characterized uniquely [17,21] by the eigenvector $\vec{x}_{\sigma}(m,k)$ of the smallest eigenvalue. The essence of the derivation is that after expansion into spherical harmonics, all the expansion coefficients can be considered as independent fluctuating densities yielding a mixture with very many "components".

We now specialize more towards our case of the ion-dipole mixture. The dimension of the matrices $\mathbf{M}(m,k)$ depends on the truncation of the expansion in equations (4.5) and (4.6). If we trucate the expansion with $l_1, l_2 \leq 3$, then $\mathbf{M}(0,k)$ is a 6×6 matrix. We introduce two new fluctuation variables

$$\delta \tilde{\rho}_{\text{ion}}(k) = \delta \tilde{\rho}_{+}(k) + \delta \tilde{\rho}_{-}(k),$$

$$\delta \tilde{\rho}_{q}(k) = \delta \tilde{\rho}_{+}(k) - \delta \tilde{\rho}_{-}(k),$$
(4.15)

which describe respectively the total ionic density fluctuation and the charge fluctuation. Using the new fluctuation variables the coefficient matrix of $\delta\Omega(0,k)$ becomes of block form of two 3×3 matrices. The total ionic density fluctuation $\delta\tilde{\rho}_{\rm ion}(k)$ couples only with $\delta\tilde{\rho}_d^{00}(k)$ and $\delta\tilde{\rho}_d^{20}(k)$. The charge fluctuation $\delta\tilde{\rho}_q(k)$ couples only with $\delta\tilde{\rho}_d^{10}(k)$ and $\delta\tilde{\rho}_d^{30}(k)$.

For the coefficient matrix related to $\delta \tilde{\rho}_{\text{ion}}(k)$, $\delta \tilde{\rho}_d^{00}(k)$ and $\delta \tilde{\rho}_d^{20}(k)$ we find the first vanishing eigenvalue at k=0, which indicates an instability with respect to homogeneous density fluctuations. In this case, $\delta \tilde{\rho}_{\text{ion}}(0)$ is only coupled with $\delta \tilde{\rho}_d^{00}(0)$ by a 2×2 matrix because the matrix elements $M_{\text{ion},(d20)}(0,k)$ and $M_{(d00),(d20)}(0,k)$ are proportional to k^2 when $k \to 0$. This 2×2 matrix later contains the smallest and finally vanishing eigenvalue λ_1 . It is convenient to define the fluctuations of total density and ionic concentration

$$\delta \bar{\rho}(0) = \rho^{-\frac{1}{2}} \left(\delta \tilde{\rho}_{\text{ion}}(0) + \delta \tilde{\rho}_{d}(0) \right),$$

$$\delta \bar{c}(0) = \rho^{-\frac{3}{2}} \left(c_{\text{ion}} c_{d} \right)^{-\frac{1}{2}} \left(\rho_{d} \delta \tilde{\rho}_{\text{ion}}(0) - \rho_{\text{ion}} \delta \tilde{\rho}_{d}(0) \right),$$
(4.16)

where $\rho = \rho_+ + \rho_- + \rho_d$, $c_{\text{ion}} = c_+ + c_-$ with $c_i = \rho_i/\rho$ and $\delta \tilde{\rho}_d(0) = (4\pi)^{\frac{1}{2}} \delta \tilde{\rho}_d^{00}(0)$. Using the new fluctuation variables the grand potential variations with respect to fluctuations $\delta \tilde{\rho}_{\text{ion}}(0)$ and $\delta \tilde{\rho}_d(0)$ can be rewritten as

$$\delta\Omega = \frac{k_{\rm B}T}{2} \frac{1}{V} \left(\delta \bar{\rho}(0) \quad \delta \bar{c}(0) \right) \begin{pmatrix} M_{\rho\rho}(0) & M_{\rho c}(0) \\ M_{c\rho}(0) & M_{cc}(0) \end{pmatrix} \begin{pmatrix} \delta \bar{\rho}(0) \\ \delta \bar{c}(0) \end{pmatrix}, \tag{4.17}$$

where the coefficient matrix $\mathbf{M}(0)$ is symmetric and defined as

$$M_{\rho\rho}(0) = 1 - \rho \left[c_{\text{ion}}^2 \tilde{c}_{\text{ion,ion}}(0) + c_d^2 \tilde{c}_{dd}^{000}(0) + 2c_{\text{ion}} c_d \tilde{c}_{+d}^{000}(0) \right], \tag{4.18}$$

$$M_{cc}(0) = 1 - \rho c_{\text{ion}} c_d \left[\tilde{c}_{\text{ion,ion}}(0) + \tilde{c}_{dd}^{000}(0) - 2\tilde{c}_{+d}^{000}(0) \right], \tag{4.19}$$

$$M_{\rho c}(0) = \rho(c_{\rm ion}c_d)^{\frac{1}{2}} \left[c_d \tilde{c}_{dd}^{000}(0) - c_{\rm ion}\tilde{c}_{\rm ion,ion}(0) - (c_d - c_{\rm ion})\tilde{c}_{+d}^{000}(0) \right]$$
(4.20)

with

$$\tilde{c}_{\text{ion,ion}}(k) = \frac{1}{2} \left[\tilde{c}_{++}(k) + \tilde{c}_{+-}(k) \right] .$$
 (4.21)

Kirkwood and Buff have related the total correlation functions to thermodynamic functions. With the help of the O.Z. equation their relations can be written into the form

$$\frac{(\rho_{\alpha}\rho_{\beta})^{\frac{1}{2}}V}{k_{\rm B}T} \left. \frac{\partial \mu_{\alpha}}{\partial N_{\beta}} \right|_{TVN'} = \mathbf{B}_{\alpha\beta}^{-1} = \delta_{\alpha\beta} - (\rho_{\alpha}\rho_{\beta})^{\frac{1}{2}} \tilde{c}_{\alpha\beta}^{000}(0), \tag{4.22}$$

where $B_{\alpha\beta} = \delta_{\alpha\beta} + (\rho_{\alpha}\rho_{\beta})^{\frac{1}{2}}\tilde{h}_{\alpha\beta}^{000}(0)$. Using the thermodynamic relations

$$\frac{\partial \mu_{\alpha}}{\partial N_{\beta}}\Big|_{T,V,N'} = \frac{\partial \mu_{\alpha}}{\partial N_{\beta}}\Big|_{T,P,N'} + \frac{\nu_{\alpha}\nu_{\beta}}{\kappa_{T}V},$$
(4.23)

$$\sum_{\alpha} N_{\alpha} \left. \frac{\partial \mu_{\alpha}}{\partial N_{\beta}} \right|_{T,P,N'} = 0 \tag{4.24}$$

with the partial volume $\nu_{\alpha} = \frac{\partial V}{\partial N_{\alpha}}\Big|_{T,P,N'}$ and the compressibility $\kappa_T = -\frac{1}{V}\frac{\partial V}{\partial P}\Big|_{T,N_i}$ the elements of the matrix $\mathbf{M}(0)$ can also be expressed by the thermodynamic functions:

$$M_{\rho\rho}(0) = (\rho k_{\rm B} T \kappa_T)^{-1}, \tag{4.25}$$

$$M_{\rho c}(0) = (c_{\rm ion}c_d)^{\frac{1}{2}}(\rho k_{\rm B}T\kappa_T)^{-1}\Delta,$$
 (4.26)

$$M_{cc}(0) = \frac{c_{\text{ion}}c_d}{V}(\rho k_{\text{B}}T)^{-1} \frac{\partial^2 G}{\partial c_{\text{ion}}^2}\Big|_{T.P.N} + c_{\text{ion}}c_d(\rho k_{\text{B}}T\kappa_T)^{-1}\Delta^2 \qquad (4.27)$$

with the difference of the partial volumes $\Delta = \rho[(\nu_+ + \nu_-)/2 - \nu_d]$ and $\partial^2 G/\partial c_{\text{ion}}^2|_{T,P,N}$ defined by

$$\frac{\partial^2 G}{\partial c_{\text{ion}}^2}\Big|_{T,P,N} = \frac{1}{2} \left(\frac{\partial^2 G}{\partial c_+^2} \Big|_{T,P,N} + \frac{\partial^2 G}{\partial c_+ \partial c_-} \Big|_{T,P,N} \right) .$$
(4.28)

After diagonalization of $\mathbf{M}(0)$ with the eigenvalues λ_i and the related eigenvectors \vec{x}_i , the $\delta\Omega(m=0,k=0)$ can be written as a sum of pure squares

$$\delta\Omega = \frac{k_{\rm B}T}{2} \frac{1}{V} [\lambda_1 |\delta \bar{\rho}_1'(0)|^2 + \lambda_2 |\delta \bar{\rho}_2'(0)|^2], \tag{4.29}$$

where

$$\delta \bar{\rho}'_i(0) = x_{i,1} \delta \bar{\rho}(0) + x_{i,2} \delta \bar{c}(0),$$
 (4.30)

$$\lambda_{1,2} = \frac{M_{\rho\rho}(0) + M_{cc}(0) \mp \sqrt{(M_{\rho\rho}(0) - M_{cc}(0))^2 + 4M_{\rho c}^2(0)}}{2} . \tag{4.31}$$

The smallest eigenvalue λ_1 decides the stability of the phase. The eigenvector \vec{x}_1 determines the softest fluctuation mode $\delta \bar{\rho}'_1(0) = x_{1,1} \delta \bar{\rho}(0) + x_{1,2} \delta \bar{c}(0)$ which has the smallest restoring force (proportional to λ_1). Therefore \vec{x}_1 characterizes properly the phase instability [17,21]. The border of a stability region is indicated by λ_1 going to zero (from the positive side). In the case $\lambda_1 = 0$ there is also Det $\mathbf{M}(0) =$ $\lambda_1 \lambda_2 = (c_{\rm ion} c_d) V^{-1} (\rho k_{\rm B} T)^{-2} \kappa_T^{-1} \partial^2 G / \partial c_{\rm ion}^2 |_{T,P,N} = 0$. There are two cases where Det $\mathbf{M}(0) = 0$. One is $\kappa_T^{-1} = 0$. In this case $\mathbf{M}(0)$ is diagonal and the softest fluctuation mode is just the total density fluctuation and the phase instability is pure condensation. Another case is $\partial^2 G/\partial c_{\text{ion}}^2|_{T,P,N}=0$. Here the softest fluctuation mode $\delta \bar{\rho}'_1(0) = \delta \bar{c}(0) - (c_{\rm ion}c_d)^{\frac{1}{2}} \Delta \delta \bar{\rho}(0)$ is generally a combination of total density and concentration fluctuation. How strong the total density will change in the softest fluctuation mode depends on the partial volume difference Δ . If $\Delta = 0$, the softest fluctuation mode is just a concentration fluctuation and the phase instability is pure demixing. When $|\Delta|$ is large, the total density will fluctuate also. The sign of Δ decides if the total density will increase or decrease. We have seen a case [21] where $\partial^2 G/\partial c^2 = 0$ and $\kappa_T^{-1} > 0$ mark an instability with respect to essentially pure total density fluctuations with negligible concentration change. Therefore there exist instabilities, where Δ is so large that $\partial^2 G/\partial c^2 = 0$ indicates a condensation instability and not a demixing.

In the following section we evaluate $\mathbf{M}(0)$ (equations (4.18)–(4.20)) from the calculated correlation functions and find a demixing instability in our electrolyte model.

5. Results for the hard sphere ion-dipole electrolyte.

We now present our calculations for the model electrolyte. The total density is fixed at the high liquid value $\rho^* = \rho \cdot \sigma^3 = 0.8$ and the ion concentration is changed by exchange of ions versus dipoles. The dipolar interaction strength is chosen as $\mu^{*2} = \mu^2/(\sigma^3 k_{\rm B}T) = 2.5$ because below $\mu^{*2} = 2.25$ the interesting missibility gap at very low ion charges and for neutral solutes does not appear [12]. The angular expansions in equation (3.4) drastically increase the dimension of the correlation function calculation (the number of independent functions). We limit ourselves to $l_1, l_2 \leq 3$, which leads to 23 unknown functions to be determined. l takes then values up to 6 due to $l \leq l_1 + l_2$. We proceed by fixing the ionic interaction $q^{*2} = q^2/(\sigma k_{\rm B}T)$ and varying the ionic concentration $c_{\rm ion} = (\rho_+ + \rho_-)/\rho$. For each system, we evaluate the correlation functions according to section 2, calculate the matrix $\mathbf{M}(k=0)$ (equations (4.18)–(4.20)) and determine the smallest of its eigenvalues λ_1 (equation (4.31)) and the associated eigenvector \vec{x}_1 . We have checked occasionally that $\lambda_1(k=0) < \lambda_i(k)$ for all k > 0.

In figure 1 we plot the smallest eigenvalue λ_1 versus $c_{\rm ion}$ for growing q^{*2} . We find two regions where the solution procedure becomes unstable and where λ_1 strongly indicates an abrupt decrease to zero. We roughly extrapolate to zero and draw the "phase-diagram" in figure 2, where the lines mean spinodales separating stable or metastable regions from unstable systems.

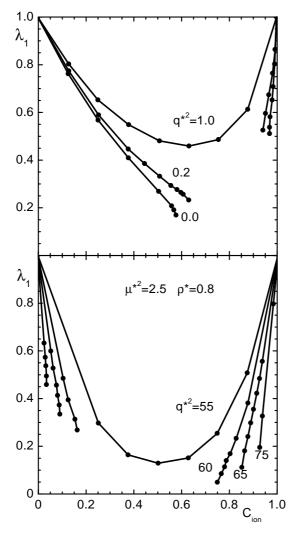


Figure 1. The smallest eigenvalue λ_1 as function of ionic concentration for different values of q^{*2} .

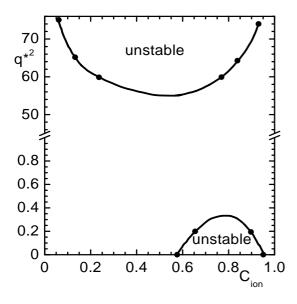
For two values of q^{*2} we also give information about the eigenvector \vec{x}_1 representing the most unstable fluctuation. We show in figure 3 the angle which this eigenvector forms in a $\delta \rho - \delta c$ -coordinate system with the positive δc axis [17,21]. The angles are small and $|\alpha| < 10^{\circ}$. This means that we find predominantly demixing instabilities.

The positive values of α for the small ion interaction $q^{*2} = 0.2$ indicate that an increase of ion concentration is accompanied by a decrease of density for the most unstable fluctuation with the smallest restoring force. This shows that the attraction between dipoles is the driving force for the instability. We have proposed in [12] that favorable dipole structures are possible for certain concentrations of neutral solutes and that this is the reason for the missibility gap. It was later pointed out to us that indications of this mechanism had been seen earlier in MC-simulations for a Stockmeyer solvent (Lenard Jones plus dipole interactions) [24].

The negative values of α for the high ionic interaction $q^{*2} = 60$ tell us that together with an increase of ionic concentration goes an increase in density, when the rise of the free energy should

be minimal. Therefore the instability is driven by the attraction of the ions. We believe that this instability is related to the "condensation" of the purely Coulombic interacting fluid found in the restricted primitive model (RPM) (see discussion).

As explained in section 2, the fluctuations of ionic concentration and of net charge density are decoupled. When the concentration fluctuations become unstable, the charge fluctuations are still very strongly constrained, i.e. the related eigenvalues of the matrix $\mathbf{M}(k)$ are very large. Homogeneous charge fluctuations (k=0) are not at all possible because the related eigenvalue is ∞ . This can be seen from $M_{q,q}(k) = 1-\rho_+\tilde{c}_{++}(k)+\rho_-\tilde{c}_{+-}(k)$ and equation (A.8) in the appendix. At large distances $c_{ij}(r)$ is proportional to the Coulomb potential and therefore its Fourier transform $\tilde{c}_{ij}(k)$ at small k goes like k^{-2} as shown in equation (A.8). Then $M_{q,q}(k) \sim k^{-2}$ goes to infinity when $k \to 0$ which suppresses homogeneous charge fluctuations completely. The total electrical neutrality of the system must be exactly satisfied.



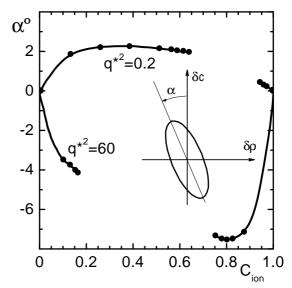


Figure 2. Regions of instablity and "spinodal" curves for the ion dipole mixture.

Figure 3. The angle α (in degrees) between the eigenvector \vec{x}_1 and the direction of pure concentration fluctuations for $q^{*2} = 0.2$ and $q^{*2} = 60$.

6. Discussion and conclusions

We are aware of several objections to the analysis of phase equilibria and stability via correlation functions calculated from integral equations. Lovett has discussed that finding the correct solution from integral equations is a "marginal" event and that approximations to the kernel, which are always necessary, should lead into the realm of no solution [25]. Belloni has demonstrated [26] that with pure HNC equations one always meets a point of bifurcation of two solutions and no solutions beyond, before the spinodal is reached where fluctuations or compressibilities become infinite or in our language the smallest eigenvalue of the matrix \mathbf{M} (equation (4.8)) goes to zero. But on the other hand experience shows that usually one gets very resonable solutions from HNC or related equations with all kind of right physical properties; and more important there is no case known to us, where the indication of a phase transition by a sudden growth of fluctuations and related impossibility of finding solutions of the integral equations by iteration was not related to a real instability of the system. This can be seen for the cases discussed by Belloni [26]. For the dipolar system we found the missibility gap with neutral solutes [12,17], the lower case in figure 1, which had been seen in simulations for dipoles and neutrals with additional Lenard-Jones interactions. The phase transition was also caused there by the dipole forces [24].

In recent years we also investigated the phase behaviour of the pure hard sphere dipole fluid. From the instability of fluctuations in the low density dipole gas we predicted the formation of aligned clusters or chains of dipoles [27]. Later on, this phase was seen in simulations [28,29]. At higher densities all the characteristics of

the unstable fluctuations could be related [30] to phases seen in simulations at lower temperature [31–33] including the transition to a ferroelectric state. This transition was of first order and the coexistence lines well framed the instability line which again confirmed the interpretation of the no-solution-line as an indication of a spinodal.

Therefore there are good reasons to expect that in the near future there will be a simulation showing the phase separation for our model electrolyte. The phase instability of the ion-dipole mixture has also been found in a study using the mean spherical approximation (MSA) for the calculation of the correlation functions [3]. This approximation underestimates the interactions and correlations and therefore finds the instability at much higher q^{*2} values or lower temperatures. It is also true that the line separating a region of no solutions from one with physically reasonable solutions of the integral equations is not simply related to the real spinodal of the system (see the discussion in 25, also [34]). Therefore the critical point or the critical behaviour cannot be safely determined via integral equations. So we consider as the main result of our investigation, that the hard sphere ion dipole mixture is unstable with respect to demixing, when the ion interactions get strong enough. We expect that for a solvent with $\mu^{*2} = 2.5$ this instability is met before the ions have one full charge. Recently there was intensive research on ionic solutions, which indeed do show a demixing phase transition [35]. With these electrolytes, one discusses the alternative of "hydrophobic" demixing or demixing due to condensation of the gas of ions towards a liquid. We have discussed in relation to figure 3 that the most unstable fluctuation indicates that the attractive forces between the ions lead to the instability, which points to the second alternative. The mechanism is then the same as in the pure Coulombic RPM of hard sphere ions. Belloni [26] shows the instability region as predicted by HNC at very low densities and q^{*2} above 7.35 ("critical point" $\rho^* = 0.3 \cdot 10^{-3}, \ q^{*2} = 7.35 \text{ or } T^* = q^{*-2} = 0.136).$ The ion dipole mixture in our RHNC treatment has the minimum of the instability curve (figure 2) near $\rho_{\text{ion}}^* = 0.4$ and $q^{*2} = 56$. When the dipole solvent is considered as a medium with dielectric constant $\epsilon = 8$, the effective interaction should be taken as $\tilde{q}^{*2} = q^2/(\epsilon \sigma k_{\rm B} T) = 7$ at the minimum. The simulations of the RPM yield the critical point at $\rho^* = 0.03$, $q^{*2} = 17.2 \ (T^* = 0.058) \ [36].$

Compared to the RPM our phase transition is found at higher densities and at higher temperatures indicating that the solvation of the ions by the solvent dipoles changes the phase transition strongly. As already seen for neutral solvents, with the ions we can also expect that special dipole configurations which require certain concentrations of ions and dipoles will play a role in the determination of the phase diagram.

Acknowledgements

We appreciate the helpful discussions with M.Kasch, M.Vossen, E.Diaz-Herrera, S.Klapp and S.Iatsevitch. We are grateful for the support via the agreement between the Max-Plank-Gesellschaft and the Chinese Academy of Science.

A. The special treatment of the long range tail of correlation functions

A.1. The direct correlation functions

The direct correlation functions have the asymptotic behaviour

$$c_{\alpha\beta}(12) \longrightarrow -u_{\alpha\beta}(12)/k_{\rm B}T, \qquad r_{12} \to \infty.$$
 (A.1)

Therefore the direct correlation function $c_{\alpha\beta}(12)$ will become long range if the pair potential $u_{\alpha\beta}(12)$ has a long range tail. In the numerical calculations the long range part of direct correlation functions needs a special treatment.

We separate the direct correlation function into two parts

$$c_{\alpha\beta}(12) = c_{\alpha\beta}^{S}(12) + c_{\alpha\beta}^{L}(12).$$
 (A.2)

 $c_{\alpha\beta}^{\rm S}(12)$ and $c_{\alpha\beta}^{\rm L}(12)$ are the short range and the long range part of $c_{\alpha\beta}(12)$, respectively. For the ion-dipole mixture we choose the long range part of the direct correlation functions in the following way

$$c_{ij}^{L}(12) = -q_i q_j e(r)/k_{\rm B}T,$$
 (A.3)

$$c_{id}^{L}(12) = -q_i \mu e'(r)(\hat{\mu}_2 \cdot \hat{r})/k_B T,$$
 (A.4)

$$c_{id}(12) = -q_i \mu e(r)(\mu_2 \cdot r)/k_B I, \qquad (A.4)$$

$$c_{dd}^L(12) = -\mu^2 [e'(r) - re''(r)] D(12)/3r/k_B T, \qquad (A.5)$$

where

$$e(r) = \frac{\operatorname{erf}(ar)}{r} \tag{A.6}$$

and e'(r) and e''(r) are the first and the second derivatives of e(r).

erf(x) in equation (1.6) is the standard error function, which is equal to zero at x=0 and approaches 1 very fast when x becomes large. Therefore the $c_{\alpha\beta}^{\rm L}(12)$ are finite at $r_{12}=0$ and possess the complete long range tail when $r_{12}\to\infty$. The separation in this way leads to continuous functions. The function e(r) was used by Ng [37] to solve the HNC equation for the classical one-component plasma in a uniform background with a=1.08. He had finished the calculation up to $q^{*2}=q^2/k_{\rm B}T\sigma=7000$ and shown the efficiency of this treatment. Now we generalize this technique to the ion-dipole mixture.

The Fourier transform of e(r) is

$$\tilde{e}(k) = \frac{4\pi}{k^2} \exp(-k^2/4a^2).$$
 (A.7)

Correspondingly the Fourier transforms of $c_{\alpha\beta}^{\rm L}(12)$ are expressed by $\tilde{e}(k)$ as

$$\tilde{c}_{ij}^{L}(k) = -q_i q_j \tilde{e}(k) / k_B T, \qquad (A.8)$$

$$\tilde{c}_{id}^{L}(12) = q_i \mu k \tilde{e}(k) (\hat{\mu}_2 \cdot \hat{k}) / k_B T, \tag{A.9}$$

$$\tilde{c}_{dd}^{L}(12) = -\mu^{2}k^{2}\tilde{e}(k)D_{k}(12)/3k_{B}T,$$
(A.10)

where $D_k(12)$ is defined by replacing \hat{r}_{12} of D(12) in equation (2.6) with the unit vector \hat{k} .

A.2. The total correlation functions

From the total correlation functions $h_{\alpha\beta}(12)$ the potential of mean force

$$W_{\alpha\beta}(12) = -k_{\rm B}T \ln[1 + h_{\alpha\beta}(12)]$$
 (A.11)

can be defined. Høye and Stell [38] have given the long range tails of $W_{\alpha\beta}(12)$ in ion-dipole mixtures

$$W_{ij}(12) = q_i q_i e_1(r),$$
 (A.12)

$$W_{id}(12) = q_i \mu_{\text{eff}} \hat{\mu}_2 \cdot \vec{\nabla}_2 e_1(r), \tag{A.13}$$

$$W_{dd}(12) = \mu_{\text{eff}}^2 \hat{\mu}_2 \cdot \vec{\nabla}_2 [\hat{\mu}_1 \cdot \vec{\nabla}_1 e_1(r)]$$
 (A.14)

with

$$e_1(r) = \frac{1}{\varepsilon |\vec{r_2} - \vec{r_1}|} \exp(-\lambda |\vec{r_2} - \vec{r_1}|),$$
 (A.15)

$$\mu_{\text{eff}} = \frac{\varepsilon - 1}{3y} \mu, \tag{A.16}$$

where ε is the dielectric constant of the fluid and $y = 4\pi \rho_{\mu} \mu^2 / 9k_{\rm B}T$. The screening parameter λ is the inverse of the Debye-Hückel length and is related to the ion charges and densities by $\lambda = \sqrt{4\pi \sum_i \rho_i q_i^2/k_{\rm B}T\varepsilon}$.

The potentials of mean force $W_{\alpha\beta}(12)$ are related to the pair potentials $u_{\alpha\beta}(12)$ in the way that $u_{\alpha\beta}(12)$ are obtained when replacing $e_1(r)$ by 1/r and μ_{eff} by μ in $W_{\alpha\beta}(12)$. Usually the inverse of the Debye-Hückel length λ is large enough that $W_{\alpha\beta}(12)$ and $h_{\alpha\beta}(12)$ decay very fast to zero when the distance is large. In some situation, for instance when the ion density is very small, the inverse of the Debye-Hückel length is small and $h_{\alpha\beta}(12)$ decay not fast. Then the long range tail of the total correlation functions $h_{\alpha\beta}(12)$ should also be treated specifically.

We divide $h_{\alpha\beta}(12)$ also into short and long range parts

$$h_{\alpha\beta}(12) = h_{\alpha\beta}^{S}(12) + h_{\alpha\beta}^{L}(12)$$
 (A.17)

Instead of the function e(r) for $c_{\alpha\beta}(12)$, the long range tails of the total correlation functions $h_{\alpha\beta}^l(12)$ are defined by function $e_2(r)$

$$e_2(r) = \frac{1}{\varepsilon r} \frac{1}{2} \left(\left[1 + \operatorname{erf}(\alpha r - \frac{\lambda}{2\alpha}) \right] \exp(-\lambda r) + \left[-1 + \operatorname{erf}(\alpha r + \frac{\lambda}{2\alpha}) \right] \exp(\lambda r) \right)$$
(A.18)

which approaches the function $e_1(r)$ for large distance and has no singularity at r = 0. The Fourier transform of the function $e_2(r)$ is

$$\tilde{e}_2(k) = \frac{1}{\varepsilon} \frac{4\pi}{k^2 + \lambda^2} \exp\left[-(k^2 + \lambda^2)/4\alpha^2\right]. \tag{A.19}$$

With the function $e_2(r)$ the long range tails of total correlation function $h_{\alpha\beta}^{L}(12)$ are defined as follows

$$h_{ij}^{L}(12) = -(q_i q_j / k_B T) e_2(r),$$
 (A.20)

$$h_{id}^{L}(12) = -(q_i \mu_{\text{eff}}/k_{\text{B}}T)e_2(r)'(\hat{\mu}_2 \cdot \hat{r}),$$
 (A.21)

$$h_{dd}^{L}(12) = -(q_{i}\mu_{\text{eff}}/\kappa_{B}T)e_{2}(T)(\mu_{2}T), \qquad (A.21)$$

$$h_{dd}^{L}(12) = \frac{\mu_{\text{eff}}^{2}}{k_{B}T} \frac{re_{2}(r)'' - e_{2}(r)'}{3r} D(12)$$

$$+ \frac{\mu_{\text{eff}}^{2}}{k_{B}T} \frac{re_{2}(r)'' + 2e_{2}(r)'}{3r} \hat{\mu}_{1} \cdot \hat{\mu}_{2} . \qquad (A.22)$$

The Fourier transforms of $h_{\alpha\beta}^{L}(12)$ have the following form

$$\tilde{h}_{ii}^{L}(12) = -(q_i q_j / k_B T) \tilde{e}_2(k),$$
(A.23)

$$\tilde{h}_{id}^{L}(12) = i(q_i \mu_{\text{eff}}/k_{\text{B}}T)k\tilde{e}_2(k)(\hat{\mu}_2 \cdot \hat{k}),$$
(A.24)

$$\tilde{h}_{dd}^{L}(12) = -\frac{\mu_{\text{eff}}^{2}}{k_{\text{B}}T}k^{2}\tilde{e}_{2}(k)(\hat{\mu}_{1}\cdot\hat{k})(\hat{\mu}_{2}\cdot\hat{k}). \tag{A.25}$$

References

- Blum L. // J. Chem. Phys., 1974, vol. 61, p. 2129; Chem. Phys. Lett., 1974, vol. 26, p. 200; J. Stat. Phys., 1978, p. 451.
- 2. Adelman A., Deutch J.M. // J. Chem. Phys., 1974, vol. 60, p. 3935.
- 3. Høye J.S., Lomba E., Stell G. // J. Chem. Phys., 1988, vol. 89, p. 7462.
- 4. Harvey A.H. // J. Chem. Phys., 1991, vol. 95, p. 479.
- 5. Levesque D., Weis J.J., Patey G.N. // J. Chem. Phys., 1980, vol. 72, p. 1887.
- 6. Fries P.H., Patey G.N. // J. Chem. Phys., 1985, vol. 82, p. 429.
- 7. Caillol J.M., Levesque D., Weis J.J. // Mol. Phys., 1990, vol. 69, p. 199.
- 8. Eggebrecht J., Ozler P. // J. Chem. Phys., 1990, vol. 93, p. 2004.
- Chan K.Y., Gubbins K.E., Henderson D., Blum L. // J. Chem. Phys., 1989, vol. 66, p. 299.
- Dong W., Rosinberg M.L., Perera A., Patey G.N. // J. Chem. Phys., 1988, vol. 89, p. 4994.
- 11. Levesque D., Weis J.J. (private communication).
- 12. Chen X.S., Kasch M., Forstmann F. // Phys. Rev. Lett., 1991, vol. 67, p. 2674.
- 13. Lado F. // Phys. Rev. A, 1964, vol. 135, p. 1013; Mol. Phys., 1976, vol. 31, p. 1117.
- Verlet L., Weis J.J. // Phys. Rev. A, 1972, vol. 5, p. 939; Henderson D., Grundke E.W. // J. Chem. Phys., 1975, vol. 63, p. 601.
- Blum L., Torruella A.J. // J. Chem. Phys., 1972, vol. 56, p. 303; ibid., 1972, vol. 57, p. 1862; 1973, vol. 58, p. 3295.
- 16. Gray C.G., Gubbins K.E. Theory of Molecular Fluids. London, Oxford, 1984, vol. 1.
- 17. Chen X.S., Forstmann F. // Mol. Phys., 1992, vol. 76, p. 1203.
- 18. Caillol J.M. // Chem. Phys. Lett., 1985, vol. 121, p. 347; 1989, vol. 156, p. 357.
- 19. Mermin N.D. // Phys. Rev., 1965, vol. 137, p. 1441.
- 20. Evans R. // Adv. Phys., 1979, vol. 28, p. 143.
- 21. Chen X.S., Forstmann F. // J. Chem. Phys., 1992, vol. 97, p. 3696.

- Lebowitz J.L., Percus J.K. // J. Math. Phys., 1963, vol. 4, p. 116; Percus J.K. The Equilibrium Theory of Classical Fluids. Ed. Frisch H.L., Lebowitz J.L., New York, Benjamin, 1964, II, p. 1–33; Saam W.F., Ebner C. // Phys. Rev. A, 1977, vol. 15, p. 2566.
- 23. Hansen J.P., McDonald I.R. Theory of Simple Liquids, London, Academic Press, 1986, 2nd Edition.
- de Leeuw S.W., Williams C.P., Smit B. // Mol. Phys., 1988, vol. 65, p. 1269; de Leeuw S.W., Smit B., Williams C.P. // J. Chem. Phys., 1990, vol. 93, p. 2704.
- 25. Lovett R. // J. Chem. Phys., 1988, vol. 88, p. 7739.
- 26. Belloni L. // J. Chem. Phys., 1993, vol. 98, p. 8080.
- 27. Kasch M., Forstmann F. // J. Chem. Phys., 1993, vol. 99, p. 3037.
- 28. Weis J.J., Levesque D. // Phys. Rev. Lett., 1993, vol. 71, p. 2729.
- 29. van Leeuwen M.E., Smit B. Poster at 2nd Liquid Matter Conf., Florence, Sept. 93; Phys. Rev. Lett., 1993, vol. 71, p. 3991.
- Klapp S., Forstmann F. // J. Chem. Phys., 1997, vol. 106, p. 9742; 1998, vol. 109, p. 1062.
- 31. Wei D., Patey G.N. // Phys. Rev. A, 1992, vol. 46, p. 7783.
- 32. Levesque D., Weis J.J. // Phys. Rev. E, 1993, vol. 48, p. 3728; 1994, vol. 49, p. 5131.
- 33. Steven M.J., Grest G.S. // Phys. Rev. E, 1995, vol. 51, p. 5976.
- 34. Fishman S., Fisher M.E. // Physica A, 1981, vol. 108, p. 1.
- 35. For a recent review see: Weingärtner H., Schröer W. // Adv. Chem. Phys., 2001, (Ed.: Prigogine I., Rice S.A.), vol. 116, p. 1.
- 36. Panagiotopoulos J.A.Z. // Fluid Phase Equilibria, 1992, vol. 76, p. 97.
- 37. Ng K.C. // J. Chem. Phys., 1974, vol. 61, p. 2680.
- 38. Høye J.S., Stell G. // J. Chem. Phys., 1978, vol. 68, p. 4145; ibid., 1979, vol. 71, p. 1985.

Фазова нестійкість незмішування в іонно-дипольній суміші

Ч.С.Шен ¹, Ф.Форстман ²

- 1 Інститут теоретичної фізики, Академія Сініка, Бейжінг, Китай
- ² Інститут теоретичної фізики, Берлінський вільний університет, Берлін, Німеччина

Отримано 2 листопада 2001 р.

Розглядаючи варіацію функціоналу великого потенціалу Ω по відношенню до малих флуктуацій густини $\delta \rho_{\alpha}$ (1), ми можемо визначити фазову нестійкість системи з кореляційних функцій. Обчислено кореляційні функції іонно-дипольної суміші, використовуючи базисне гіперланцюгове наближення. На основі отриманих кореляційних функцій була досліджена фазова поведінка іонно-дипольних сумішей.

Ключові слова: фазовий перехід, незмішування, нестійкість, іонно-дипольна суміш, інтегральні рівняння

PACS: 61.20.Gy, 64.60.-i, 64.75.+g, 64.70.-p, 64.30.+t, 61.20.Q