

Fig. 1. X-ray diffraction curves for (200) reflection of SL EuS-PbSe with a period 20 nm as-deposited (a) and after annealing at temperature 733 K within 0.5 h (b) and 1.0 h (c). S_k — satellites.

sity but also to determine their interdiffusion coefficients [11, 12]:

$$\ln[I_k(\tau_2)/I_k(\tau_1)] = -8k^2\pi^2D(\tau_2 - \tau_1)/H^2, \quad (1)$$

where D — diffusion coefficient; H — period of SL; k — order of satellite; I_k — relative intensity of k -th satellite normalized on zero satellite intensity; τ — annealing time.

The subjects of investigation in this work are EuS-PbS, EuS-PbSe, EuS-SrS and EuS-PbTe SL's which were made in oil-free vacuum (10^{-4} – 10^{-5} Pa) by thermal evaporation of lead chalcogenides from tungsten boats and electron-beam evaporation of EuS and their consecutive condensation onto (001) KCl at 523 K. Layer thickness and condensation speed were supervised by the calibrated quartz resonator located near a substrate. The parameters of prepared SL's are represented in Table 1. The samples were separated from the substrates by dissolution KCl in distillation water and were transferred onto object glasses. The structural researches (electron microscopy and X-ray diffraction) have shown that SL's have the monocrystalline

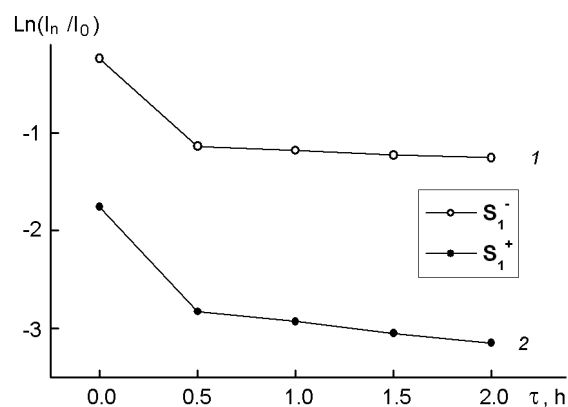


Fig. 2. Variation of satellite relative intensity for the first S_1^- (1) and S_1^+ (2) order due to annealing time τ at 733 K for SL EuS-PbSe with period of 20 nm.

layers in (001) orientation with sharp interfaces. This indicated by the presence of satellites of the 2-rd order in the X-ray diffractions (Fig. 1a). The samples were annealed in vacuum at three different temperatures for each SL's (see Table 2). The samples were taken out from the vacuum chamber periodically for X-ray diffraction examinations. Diffraction curves were taken using a double-crystal spectrometer in $\text{Cu-K}\alpha_1$ radiation and $\varphi - 2\varphi$ scanning regime ((400) reflection of the silicon monochromator crystal was used). In Fig. 1 it is seen that during an annealing of SL EuS-PbSe the intensity of lateral satellites decreases and central (zero) satellite increases indicating intermixing of the layers. More evidently process of layer intermixing in SL EuS-PbSe at 733 K is submitted in Fig. 2, from which it is visible, that there are two stages of diffusion — the fast (on the initial stages of annealing) and the slow one. The fast diffusion on the initial stages of annealing is evidently due to the presence of nonequilibrium nonstoichiometric point defects in lead chalcogenides [13].

The similar dependences of satellite intensity changes are observed at temperatures 593 K and 693 K.

Table 1. Parameters of the samples: f , is the lattice misfit of layers; N , number of layers; H , the SL period.

SL	f , %	N	H , nm
EuS-PbS	0.5	20	8.0; 12.0; 18.0
EuS-PbSe	2.5	20	11.6; 19.0; 20.0
EuS-PbTe	7.7	20	10.2; 12.0; 16.2
EuS-SrS	0.8	20	10.0; 12.4; 16.0

Table 2. Diffusion characteristics of SLs: H , the SL period; T , the annealing temperature; D , diffusion coefficient; D_0 , pre-exponential factor; E_a , activation energy; $D_{523\text{ K}}$, diffusion coefficient at 523 K; ΔX , the intermixed zone thickness at 523 K for 1 h

CL	H , nm	T , K	D , cm ² /s	D_0 , cm ² /s	E_a , eV	$D_{523\text{ K}}$, cm ² /s	ΔX , nm
EuS–PbS	8.0	543	$1.1 \cdot 10^{-20}$	$2.2 \cdot 10^{-9}$	1.22	$3 \cdot 10^{-21}$	0.06
	12.0	593	$6.4 \cdot 10^{-20}$				
	18.0	623	$1.7 \cdot 10^{-18}$				
EuS–PbSe	19.0	593	$7.7 \cdot 10^{-21}$	$4.0 \cdot 10^{-10}$	1.26	$2.9 \cdot 10^{-22}$	0.02
	11.6	693	$3.3 \cdot 10^{-19}$				
	20.0	733	$8.7 \cdot 10^{-19}$				
EuS–SrS	12.4	673	$6.2 \cdot 10^{-21}$	$1.7 \cdot 10^{-8}$	1.3	$3.2 \cdot 10^{-19}$	0.7
	10.0	723	$4.5 \cdot 10^{-20}$				
	16.0	773	$1.1 \cdot 10^{-18}$				
EuS–PbTe	Are not intermixed up to 800 K						

The layer interdiffusion coefficients for SL EuS–PbSe were determined for each temperatures using expression (1):

$$D = 7.7 \cdot 10^{-21} \text{ cm}^2/\text{s} \text{ (593 K);}$$

$$3.3 \cdot 10^{-19} \text{ cm}^2/\text{s} \text{ (693 K);}$$

$$8.7 \cdot 10^{-19} \text{ cm}^2/\text{s} \text{ (733 K).}$$

Taking into account the Arrhenius law ($D = D_0 \exp(E_a/kT)$), the values of activation energy E_a and pre-exponential coefficient D_0 were determined for SL EuS–PbSe:

$$D_0 = 4.0 \cdot 10^{-10} \text{ cm}^2/\text{s}; \quad E_a = 1.26 \text{ eV.}$$

Thus it is possible to calculate the diffusion coefficient for any necessary temperature (for-example, for SL growth temperature 523 K) and to estimate the intermixing zone ΔX formed during SL growth time ($t \sim 1$ h):

$$\Delta X^2 \sim 4Dt. \quad (2)$$

The results of such calculations for SL EuS–PbSe are $D_{523\text{ K}} = 2.9 \cdot 10^{-22} \text{ cm}^2/\text{s}$; $\Delta X = 0.02 \text{ nm}$.

The same procedure was performed for SL EuS–PbS and EuS–SrS and the results of investigations are presented in Table 2.

It should be noted that layer materials in SL EuS–PbTe are not intermixed up to 800 K until re-evaporation of materials was observed.

The results are presented in Table 2. Basing of the results the SLs studied are considered to be rather stable structures having in as-prepared state very small intermixed zone thickness of the order of one monolayer or less, thus being very promising objects for further studies and applications.

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